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# Novel A－seco－nortriterpenoids from Ganoderma cochlear inhibiting Tau pathology by activating AMPK－ULK1－mediated autophagy $\dagger$ 

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#### Abstract

Ten structurally diverse Ganoderma triterpenoids，including one unprecedented highly degraded A－seco－ nortriterpenoid，ganolearin $A(1)$ with a $6 / 6 / 5$－tricyclo structure containing a unique benzene ring，three new rearranged nortriterpenoids，ganolearin $B(2)$ and ganolearin $C$（3）with a rare $3 / 5 / 6 / 5$－fused skeleton， and ganolearin $D(4)$ featuring a $3 \rightarrow 10 \gamma$－lactone ring and a five－membered carbon ring，and six analogues， ganolearates $\mathrm{E}-\mathrm{G}$ and $\mathrm{J}(\mathbf{5}-\mathbf{7}, \mathbf{1 0})$ ，and ganolearic acids $\mathrm{H}(\mathbf{8})$ and $\mathrm{I}(9)$ ，were isolated from the fruiting bodies of Ganoderma cochlear．Their structures were elucidated by extensive 1D and 2D NMR spectroscopy， HRESIMS，X－ray crystallography and ECD calculation analysis．A plausible biosynthetic pathway for 1－4 was proposed．Furthermore，compounds 1－4 significantly inhibited Tau pathology by inducing autophagy mediated by the AMPK－ULK1 pathway，suggesting their potential against Alzheimer＇s disease（AD）．


## Introduction

Ganoderma has attracted significant attention from pharmacol－ ogists，chemists，and phytochemists because it has long been used as a traditional medicine in many countries ${ }^{1,2}$ and is also an important source of meroterpenoids and lanostane triter－ penoids with significant bioactivities and captivating structures．${ }^{3-9}$ In particular，Ganoderma triterpenoids（GTs） have been widely studied as the main active constituents of Ganoderma for their anti－tumor，anti－aging，liver－protective，

[^0]and neuroprotective activities．Notably，previous studies have confirmed that GTs can reverse cognitive impairment，${ }^{10}$ attenuate LPS－induced inflammation and apoptosis，${ }^{11}$ promote amyloid－$\beta$ clearance and inhibit Tau pathology through activating autophagy，${ }^{6,12}$ suggesting that GTs exhibit huge potential in the treatment of neurodegenerative diseases．

Alzheimer＇s disease（AD），the most pervasive neurodegen－ erative disease associated with aging，seriously threatens people＇s lives．${ }^{13,14}$ The extracellular amyloid plaques（which are composed of amyloid beta（ $\mathrm{A} \beta$ ）peptides）and the intra－ neuronal neurofibrillary tangles（NFTs）composed of the Tau （MAPT／Tau）protein are two hallmarks of AD pathology． Autophagy is a key process wherein cellular components such as corrupted organelles or unwanted protein materials aggre－ gate in intracellular autophagosomes and undergo degra－ dation or／and recycling in lysosomes．${ }^{15}$ Various studies prove that autophagy is the main route to remove $\mathrm{A} \beta$ or／and MAPT／ Tau aggregates．${ }^{16,17}$ Thus，activation of autophagy might be a vital strategy for AD treatment．${ }^{18-22}$

Therefore，we continue isolating structurally diverse GTs from the fruiting bodies of Ganoderma cochlear in order to provide more possibilities for the development of anti－AD drugs．Interestingly，ten new A－seco－norlanostane triterpenoids with five different skeletons were found（Fig． 1 and Fig． $\mathrm{S} 1 \dagger$ ）． Their structures were confirmed by using 1D and 2D NMR spec－ troscopic data，X－ray single crystollography，and electronic circu－ lar dichroism（ECD）calculation methods．The further cellular assay indicated that compounds 1－4 induced autophagy by acti－ vating the AMPK－ULK1 signaling pathway（mTOR－independent



Fig. 1 Structures of compounds 1-4.
pathway). Overall, the current data demonstrated that 1-4 could be potential compounds for AD therapy by inducing autophagy.

## Results and discussion

Ganolearin A (1) has a molecular formula of $\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{5}$, as determined by HRESIMS at $m / z 401.2322[\mathrm{M}+\mathrm{H}]^{+}$(calcd 401.2323), indicating 9 degrees of unsaturation. The ${ }^{1} \mathrm{H}$ NMR spectroscopic data (Table $\mathrm{S} 1 \dagger$ ) of $\mathbf{1}$ showed four singlet methyls at $\delta_{\mathrm{H}} 0.74,1.21,2.55$, and 2.57 , one doublet methyl at $\delta_{\mathrm{H}} 0.90(\mathrm{~d}, J=6.4 \mathrm{~Hz})$, one methoxyl at $\delta_{\mathrm{H}} 3.68(\mathrm{~s})$, two aromatic methines at $\delta_{\mathrm{H}} 7.47(\mathrm{~d}, J=7.9 \mathrm{~Hz})$ and $\delta_{\mathrm{H}} 7.54(\mathrm{~d}, J=7.9$ Hz ), and one oxygenated methine at $\delta_{\mathrm{H}} 5.60(\mathrm{dd}, J=5.1$ and 9.5 Hz ). In the ${ }^{13} \mathrm{C}$-DEPT spectra of $\mathbf{1}$, twenty-four carbon resonances were observed, comprising five methyls, one methoxyl, four methylenes, five methines, and nine quaternary carbons (two ketones, one ester carbonyl, and four aromatic quaternary carbons). The above data showed that compound 1 could be a norlanostane triterpenoid.

The HMBC correlations (Fig. 2) of $\mathrm{H}_{3}-18$ with $\mathrm{C}-12, \mathrm{C}-13$, $\mathrm{C}-14$, and $\mathrm{C}-17$; of $\mathrm{H}_{2}-12$ with $\mathrm{C}-11\left(\delta_{\mathrm{C}} 200.4\right)$ and $\mathrm{C}-9\left(\delta_{\mathrm{C}}\right.$ 140.7); of $\mathrm{H}_{3}-21$ with $\mathrm{C}-17, \mathrm{C}-20$, and $\mathrm{C}-22$; and of $\mathrm{H}_{2}-22$ and OMe ( $\delta_{\mathrm{H}} 3.68$, s) with $\mathrm{C}-24\left(\delta_{\mathrm{c}} 174.2\right.$ ), as well as a series of ${ }^{1} \mathrm{H}^{1}{ }^{1} \mathrm{H}$ COSY correlations (Fig. 2) of $\mathrm{H}-15 / \mathrm{H}_{2}-16 / \mathrm{H}-17 / \mathrm{H}-20 / \mathrm{H}_{2}-$ $22 / \mathrm{H}_{2}-23$, indicated that compound 1 had similar C and D


Fig. 2 Selected HMBC ( $\mathrm{H} \rightarrow \mathrm{C}$ ) and ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H} \operatorname{COSY}(\mathrm{H}-\mathrm{H})$ correlations of compounds 1-4.
rings and side chains to those of fornicatin E. ${ }^{23}$ Further analysis of the HMBC spectrum (Fig. 2) showed that $\mathrm{H}_{3}-30$ correlated with $\mathrm{C}-13, \mathrm{C}-14, \mathrm{C}-15\left(\delta_{\mathrm{C}} 74.1\right)$, and $\mathrm{C}-8\left(\delta_{\mathrm{C}} 152.8\right)$; simultaneously, $\mathrm{H}-15$ showed correlations with $\mathrm{C}-13, \mathrm{C}-8$, and $\mathrm{C}-17$. The aforementioned information illustrated that the hydroxyl group was linked to $\mathrm{C}-15$. These functionalities accounted for five degrees of unsaturation.

Except for a ketone carbonyl, the remaining 4 degrees of unsaturation were representative of another benzene ring, which was consistent with the observation of six aromatic signals at $\delta_{\mathrm{H}} 7.47(\mathrm{~d}, J=7.9 \mathrm{~Hz}), \delta_{\mathrm{C}} 122.6 ; \delta_{\mathrm{H}} 7.54(\mathrm{~d}, J=7.9$ $\mathrm{Hz}), \delta_{\mathrm{C}} 131.2 ; \delta_{\mathrm{C}} 131.2 ; \delta_{\mathrm{C}} 138.1 ; \delta_{\mathrm{C}} 140.7$ and $\delta_{\mathrm{C}}$ 152.8. The HMBC correlations (Fig. 2) of $\mathrm{H}_{3}-19$ with C-5 ( $\delta_{\mathrm{C}} 131.2$ ), C-9 ( $\delta_{\mathrm{C}}$ 140.7), and $\mathrm{C}-10$ ( $\delta_{\mathrm{C}} 138.1$ ), of $\mathrm{H}_{3}-29$ with $\mathrm{C}-5$ and $\mathrm{C}-4$ ( $\delta_{\mathrm{C}}$ 203.9), of H-6 ( $\delta_{\mathrm{H}} 7.54, \mathrm{~d}, J=7.9 \mathrm{~Hz}$ ) with C-4, C-10, and C-8 ( $\delta_{\mathrm{C}} 152.8$ ), and of $\mathrm{H}-7$ ( $\delta_{\mathrm{H}} 7.47, \mathrm{~d}, J=7.9 \mathrm{~Hz}$ ) with C-5, C-9 ( $\delta_{\mathrm{C}}$ 140.7 ), and $\mathrm{C}-14\left(\delta_{\mathrm{C}} 52.3\right)$ further confirmed that $\mathrm{C}-1, \mathrm{C}-2, \mathrm{C}-3$, and $\mathrm{C}-28$ were degraded and the B ring was aromatized. In the ROESY spectrum (Fig. 3) of 1, H-15 correlated with $\mathrm{H}_{3}-18$, indicating that $15-\mathrm{OH}$ was $\alpha$-oriented. Therefore, the structure of compound 1 was finally established.

The molecular formula of ganolearin B (2) was determined to be $\mathrm{C}_{26} \mathrm{H}_{38} \mathrm{O}_{6}$ based on the positive HRESIMS at $\mathrm{m} / \mathrm{z}$ $469.2564[\mathrm{M}+\mathrm{Na}]^{+}$(calcd 469.2561), which indicated 8 degrees of unsaturation. The 1D NMR spectroscopic data (Table S1 $\dagger$ ) of 2 showed that the structure of 2 resembles that of ganolearic $\operatorname{acid} \mathrm{A},{ }^{24}$ except for the presence of an additional methoxyl and an oxymethine, and the absence of one methylene. The detailed comparison of the 1D NMR spectra of 2 and ganolearic acid A showed the replacement of the carboxyl at C-24 by a methyl ester carbonyl, which was confirmed by the HMBC correlations (Fig. 2) of $\mathrm{OCH}_{3}\left(\delta_{\mathrm{H}} 3.71\right.$, s) with $\mathrm{C}-24\left(\delta_{\mathrm{C}} 176.2\right)$ in 2. Furthermore, the oxymethine proton ( $\delta_{\mathrm{H}} 4.71$, dd, $J=12.9$ and 5.9 Hz ) showed HMBC correlations with C-13 ( $\delta_{\mathrm{C}} 48.1$ ), C-14 ( $\delta_{\mathrm{C}} 48.2$ ), C-15 ( $\delta_{\mathrm{C}} 40.9$ ), and $\mathrm{C}-17\left(\delta_{\mathrm{C}} 54.7\right)$, as well as ${ }^{1} \mathrm{H}^{1}{ }^{1} \mathrm{H}$ COSY correlations (Fig. 2) of $\mathrm{H}_{2}-15 / \mathrm{H}-16 / \mathrm{H}-17 / \mathrm{H}-20 / \mathrm{H}_{3}-21$, suggesting that C-15 was connected to a hydroxyl group. Thus, the planar structure of 2 was established.

The ROESY spectrum of 2 showed the correlations of $\mathrm{H}-16 /$ $\mathrm{H}_{3}-30$, indicating that $16-\mathrm{OH}$ was $\beta$-oriented (Fig. 3). In


3

2

4

Fig. 3 Key ROESY (, - ) correlations of compounds 1-4.
addition, the ROESY correlations of $\mathrm{H}_{3}-19 / \mathrm{H}-5 / \mathrm{H}-7 / \mathrm{H}-6 \mathrm{a}$ and of $\mathrm{H}-6 \mathrm{~b} / \mathrm{H}_{3}-30$ were observed, indicating that the relative configurations of $\mathrm{H}-5$ and $\mathrm{H}-7$ were $\beta$, opposite to $5 \alpha$ and $7 \alpha$ of ganolearic acid A. ${ }^{24}$ Furthermore, the X-ray crystallographic data (Fig. 4A) proved that the absolute configuration of 2 was $5 S, 7 R, 10 S, 13 R, 14 R, 16 S, 17 R, 20 R$. Thus, the structure of compound 2 including the stereostructure was finally determined.

The HRESIMS of ganolearin C (3) showed an $[\mathrm{M}-\mathrm{H}]^{-}$peak at $m / z 457.2602[\mathrm{M}-\mathrm{H}]^{-}$(calcd 457.2596), indicating a molecular formula of $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{O}_{6}$ for 3 and 9 degrees of unsaturation. Its 1D NMR spectroscopic data (Table $\mathrm{S} 1 \dagger$ ) were similar to those of ganolearic acid $\mathrm{A},{ }^{24}$ suggesting that compound 3 was also a norlanostane triterpenoid with a 3/4/6/4-tetracyclic skel-
A

B


Fig. 4 (A) X-ray crystallographic structures of compounds 2 and 4. (B) Calculated and experimental CD curves of compound 3 .
eton. However, an additional methyl group ( $\delta_{\mathrm{H}} 2.25, \mathrm{~s} ; \delta_{\mathrm{C}} 30.5$ ) and a ketone group ( $\delta_{\mathrm{C}} 205.4$ ) were present in 3; meanwhile, one methylene in ganolearic acid A was replaced by a methine ( $\delta_{\mathrm{H}} 1.62, \mathrm{~m} ; \delta_{\mathrm{C}} 35.7$ ) in 3. Furthermore, the HMBC spectrum (Fig. 2) of 3 showed the correlations of the methyl protons ( $\delta_{\mathrm{H}}$ 2.25 , s) with the ketone carbonyl ( $\delta_{\mathrm{C}} 205.4$ ) and methine ( $\delta_{\mathrm{C}}$ 35.7), of the methine proton ( $\delta_{\mathrm{H}} 1.62, \mathrm{~m}$ ) with C-6 ( $\delta_{\mathrm{C}} 39.2$ ), C-10 ( $\delta_{\mathrm{C}} 49.7$ ), and C-8 ( $\delta_{\mathrm{C}} 171.8$ ), of H-6 ( $\delta_{\mathrm{H}} 2.04, \mathrm{~m}$ ) with C-10 ( $\delta_{\mathrm{C}} 49.7$ ), C-9 ( $\delta_{\mathrm{C}} 135.6$ ), and $\mathrm{C}-19\left(\delta_{\mathrm{C}} 22.5\right)$, and of $\mathrm{H}-7\left(\delta_{\mathrm{H}}\right.$ $2.47, \mathrm{~m}$ ) with $\mathrm{C}-9$ and $\mathrm{C}-14\left(\delta_{\mathrm{C}} 49.7\right)$. Meanwhile, the ${ }^{1} \mathrm{H}^{1}{ }^{1} \mathrm{H}$ COSY correlations (Fig. 2) of $\mathrm{H}-5 / \mathrm{H}-6 / \mathrm{H}-7$ illustrated that $\mathrm{C}-5$ was linked with an acetyl group. The ROESY correlations (Fig. 3) of $\mathrm{H}-5 / \mathrm{H}_{3}-18 / \mathrm{H}_{3}-19$ and $\mathrm{H}-6 / \mathrm{H}-7 / \mathrm{H}_{3}-30$ indicated that $\mathrm{H}-5$ was $\beta$-oriented; in contrast, H-6 and H-7 were $\alpha$-oriented. Furthermore, the ECD calculation method was used to confirm its absolute configuration. As shown in Fig. 4B, the ECD curve of $5 R, 6 S, 7 R, 10 S, 13 R, 14 R, 17 R, 20 R-3$ corresponds to the experimental CD curve. Therefore, the structure of 3 was established.

Ganolearin D (4) has a molecular formula of $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{6}$, which was determined using HRESIMS at $m / z 471.2143$ [ $\mathrm{M}+$ $K]^{+}$(calcd 471.2143) with 8 degrees of unsaturation. It had similar 1D NMR spectroscopic data to those of ganocochlearic acid $\mathrm{A}^{25}$ except that an oxymethine replaced the methylene in 4. Furthermore, the oxymethine proton ( $\delta_{\mathrm{H}} 5.36, \mathrm{t}, J=7.3 \mathrm{~Hz}$ ) showed the HMBC correlations of C-5 ( $\delta_{\mathrm{C}} 48.5$ ), C-6 ( $\delta_{\mathrm{C}} 38.5$ ), C-8 ( $\delta_{\mathrm{C}} 174.1$ ), C-9 ( $\delta_{\mathrm{C}} 133.7$ ), and $\mathrm{C}-14\left(\delta_{\mathrm{C}} 49.8\right)$, which indicated that the hydroxyl group was located at C-7. The ROESY correlation of $\mathrm{H}-7 / \mathrm{H}_{3}-30$ proved the $\beta$-configuration of $7-\mathrm{OH}$. The X-ray crystallographic data (Fig. 4A) further confirmed that the absolute configuration of 4 was $5 S, 7 S, 10 R, 13 R, 14 R, 17 R, 20 R$. Finally, the structure of 4 was determined.

Compounds 1-4 were highly degraded and rearranged lanostane-triterpenoids, in which major changes were present in the B ring. Biosynthetically, fornicatin $\mathrm{A}^{23}$ was deduced to be a plausible biosynthetic precursor, which was further oxidized to a 4-oxo derivative (intermediate I). Subsequently, the


Scheme 1 A plausible biosynthetic pathway for 1-4.


Fig. 5 Inhibition effects of compounds 1-4 on Tau pathology by activating autophagy in SH-SY5Y MAPT cells. (A-H) Western blotting assays showing the protein levels of autophagy markers LC3-II/LC3-I and SQSTM1, and Tau P301S in the SH-SY5Y MAPT cells treated with or without the compounds. (I, K) Western blotting analyses of autophagy markers LC3-II/LC3-I and SQSTM1, and Tau P301S in cell lysates from SH-SY5Y MAPT cells treated with BAFA1 (bafilomycin A1, 20 nM ), 1-4 (20 $\mu \mathrm{M}$ ) or both (BAFA1 and 1, BAFA1 and 2, BAFA1 and 3, or BAFA1 and 4). (A-L) A representative western blotting result ( $\mathrm{A}, \mathrm{C}, \mathrm{E}, \mathrm{G}, \mathrm{I}$ and K ) and quantification of the respective protein levels ( $\mathrm{B}, \mathrm{D}, \mathrm{F}, \mathrm{H}, \mathrm{J}$ and L ) based on 3 independent experiments. DMSO (dimethyl sulfoxide), used as a solvent for the compounds, served as the control. Rapamycin (Rapa) was used as a positive control. Relative protein abundance was normalized to GAPDH. ns, not significant; *, $P<0.05$; **, $P<0.01$; ***, $P<0.001$; ****, $P<0.0001$; one-way ANOVA with Tukey's post-hoc test. Bars represent mean $\pm$ SD.
dehydration of 7-OH and H-6 happened, forming II containing a long-range unsaturated fraction (C-6/C-7/C-8/C-9/C-11). The allylic oxidation at $\mathrm{C}-5$ led to the formation of III, which was dehydrated to form a carbocation (IV). The further aromatization of ring B occurred by losing the $\mathrm{C}-1 / \mathrm{C}-2 / \mathrm{C}-3$ moiety. The oxidation of $\mathrm{C}-15$ and the esterification of $24-\mathrm{COOH}$ finally resulted in the formation of compound 1. In addition, II can also be transformed to $\mathbf{V}$ via a 1,2-alkyl shift under acidic conditions. The cyclization between C-7 and C-5 and esterification at $\mathrm{C}-3$ resulted in the formation of 3 .

VI was obtained from I through a Baeyer-Villiger reaction. Then, the hydrolyzation of VI led to the formation of VII and the carbocation at C-5 was obtained as VIII through the dehydration of $5-\mathrm{OH}$ because of the acid. Meanwhile, the cleavage of $\mathrm{C}-10 / \mathrm{C}-9$ and the linkage of $\mathrm{C}-9 / \mathrm{C}-5$ resulted in the formation of a carbocation at C-10. Simultaneously, 3-COOH attacked $\mathrm{C}-10$ to form $3 \rightarrow 10 \gamma$-lactone (compound 4). Additionally, the dehydration of $7-\mathrm{OH}$ occurred under the acidic conditions, forming a carbocation (C-7). C-5 further
attacked C-7, resulting in the formation of a unique threemembered ring accompanied by the deacetyloxy group. Both $3-\mathrm{COOH}$ and $24-\mathrm{COOH}$ were finally esterified to obtain compound 2 (Scheme 1).

It is well known that autophagy is a promising therapeutic strategy against tauopathy for promoting the clearance of Tau. ${ }^{26}$ In our present study, compounds $\mathbf{1 - 4}$ could significantly induce autophagy and show no toxicities in SH-SY5Y mCherry-GFP-LC3 cells, at concentrations of 5 and $20 \mu \mathrm{M}$ (Fig. S4 and $\mathrm{S} 5 \dagger$ ). Then, the human MAPT mutant MAPT-p.P301S was stably expressed in SH-SY5Y cells (named SH-SY5Y MAPT cells). ${ }^{6}$ In the SH-SY5Y MAPT cells treated with $\mathbf{1 - 4}$, an increased protein level of the lipidated (PE-conjugated) form of MAP1LC3/LC3 (microtubule-associated protein 1 light chain 3; LC3-II/LC3-I) and a decreased protein level of SQSTM1 (sequestosome 1) in a dose-dependent manner were observed (Fig. 5A-H).

Meanwhile, compounds 1-4 reduced the Tau P301S level in SH-SY5Y MAPT cells (Fig. 5A-H). Furthermore, BAFA1 (bafilo-


Fig. 6 Compounds 1-4 activate the AMPK-ULK1 pathway. (A-H) Western blotting assays showing the protein levels of pAMPK $\alpha$, AMPK $\alpha$, pAMPK $\beta 1$, AMPK 31, AMPK $\beta 2$, pULK1ser777, pULK1ser757, pULK1ser555 and ULK1 in the SH-SY5Y MAPT cells treated with or without the compounds. (A-H) A representative western blotting result ( $A, C, E$ and $G$ ) and quantification of the respective protein levels ( $B, D, F$ and $H$ ) based on 3 independent experiments. Relative protein abundance was normalized to ACTB. ns, not significant; *, $P<0.05 ; * *, P<0.01$; ***, $P<0.001$; one-way ANOVA with Tukey's post-hoc test. Bars represent mean $\pm$ SD.


Fig. 7 The molecular docking results of compounds 1-4 with AMPK (PDB: 4CFH).
mycin A1), an inhibitor of the vacuolar(v)-type ATPase resulting in the blockage of autophagosome-lysosome fusion and accumulation of LC3-II, ${ }^{27}$ was further used to demonstrate the role of autophagy induced by 1-4 in MAPT/Tau clearance in SH-SY5Y MAPT cells. Treatment with BAFA1 ( 20 nM ) alone raised LC3-II/LC3-I, but with no significant effect on Tau P301S. BAFA1 could reverse the decreased levels of Tau P301S induced by compounds 1-4 (Fig. 5I-L). These results illustrated that compounds 1-4 reduce Tau P301S by activating autophagy.

Furthermore, we investigated the mechanism by which compounds 1-4 regulate autophagy. First, we checked whether these compounds would affect the mTOR activity. Rapamycin (an inhibitor of mTOR) is used as a positive control. Our results show that the activity of mTOR is not influenced by compounds 1-4 because of no significant changes in the protein levels of mTOR, pmTOR2448 and pmTOR2481 in SH-SY5Y MAPT cells (Fig. S6 $\dagger$ ).

As AMPK activation is one of the critical pathways involved in autophagy induction, ${ }^{28}$ recent studies have reported that the inactivation of mTOR is not always necessary for autophagy. ${ }^{29}$ AMPK can also directly phosphorylate and activate ULK1 at Ser317, Ser555, and Ser777 to initiate autophagy. ${ }^{30}$ In the present study, compounds $\mathbf{1 - 4}$ significantly increased the protein levels of pAMPK $\alpha$, pAMPK $\beta 1$, pULKser777, pULKser317, and pULKser555, whereas the protein levels of AMPK $\alpha$, AMPK $\beta 1$, AMPK $\beta 2$, ULK1 and pULKser757 did not significantly change after treatment with compounds $\mathbf{1 - 4}$ in SH-SH5Y MAPT cells (Fig. 6). The further molecular docking revealed the interaction mode of compounds 1-4 with AMPK, and their binding energy ranged from -7.7 to $-8.2 \mathrm{kcal} \mathrm{mol}^{-1}$. Compounds 1-4 bound the key domain of AMPK containing ARG263 (Fig. 7), which affects AMPK phosphorylation. These results suggest that compounds $\mathbf{1 - 4}$ induce autophagy by activating the AMPK-ULK1 pathway rather than the mTOR-dependent signaling pathway.

## Conclusions

In summary, we isolated ten A-seco-nortriterpenoids with six different types of skeletons from G. cochlear, of which com-
pound $\mathbf{1}$ is the first example of a 6/6/5-tricyclic-hexanortriterpenoid containing a unique aromatic ring. Although 2 and 3 have similar skeletons, combined analysis of their detailed structures and stereochemistry suggests that they could be derived via different biosynthetic pathways. The autophagy process is mainly regulated through mTOR-dependent and mTOR-independent signaling pathways. ${ }^{28}$ Previous research studies showed that GTs reverse cognitive impairment by suppressing PI3K/AKT/mTOR expression. ${ }^{10}$ However, in our study, compounds 1-4 could significantly reduce Tau pathology by activating autophagy mediated by the AMPK-ULK1 pathway rather than the mTOR-dependent pathway, indicating that GTs with different skeletons can exhibit anti-neurodegeneration via multiple targets and signaling pathways. Meanwhile, these results also provide scientific evidence for the application of Ganoderma in anti-neurodegenerative diseases.

## Experimental section

## General experimental procedures

Silica gel (200-300 mesh, Qingdao Marine Chemical, Inc.), Lichroprep RP-18 (40-63 $\mu \mathrm{m}$, Fuji) and Sephadex LH-20 (20-150 $\mu \mathrm{m}$, Pharmacia) were used for column chromatography. Methanol, chloroform, ethyl acetate, acetone, petroleum ether, $n$-hexane and 2-propanol were purchased from Tianjin Chemical Reagents Co. (Tianjin, China). A Shimadzu UV-2401PC spectrometer was used for recording ultraviolet (UV) spectra. A Horiba SEPA-300 polarimeter was used for optical rotations. A Chirascan instrument was used for recording CD spectra. A Bruker AV-600 MHz spectrometer (Bruker, Zurich, Switzerland) was used for recording nuclear magnetic resonance (NMR) spectra and tetramethyl chlorosilane (TMS) was used as an internal standard for chemical shifts. Electrospray ionization mass spectrometry (ESIMS) and HRTOF-ESIMS spectra were recorded using an API QSTAR Pulsar spectrometer. A Bruker Tensor-27 instrument with KBr pellets was used for recording IR spectra. An Agilent 1100 series instrument equipped with an Agilent ZORBAX SB-C18 column ( $5 \mu \mathrm{~m}, 9.6 \mathrm{~mm} \times 250 \mathrm{~mm}$ ) was used for high-performance liquid chromatography (HPLC) separation.

## Fungal materials

Ganoderma cochlear samples ( 44 kg ) were purchased in December 2020 from the Traditional Chinese Medicine Market in Kunming, Yunnan, China, which was identified by Prof. Yang Zhuliang, Kunming Institute of Botany, Chinese Academy of Sciences (voucher no. 20202501).

## Extraction and isolation

G. cochlear samples ( 44 kg ) were extracted three times with $95 \%$ ethanol $(\mathrm{EtOH})(80 \mathrm{~L} \times 3)$ under reflux ( 3 h per extraction). The combined ethanol extracts were evaporated under reduced pressure. The residue ( 4 kg ) was suspended in $\mathrm{H}_{2} \mathrm{O}$ and extracted with ethyl acetate (EtOAc). The volume of the combined EtOAc extracts ( 1.5 kg ) was reduced to one-third under
reduced pressure. The residue was fractionated by using a silica gel column $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}, 100: 0,80: 1,50: 1,20: 1\right.$, $10: 1$, and $5: 1, \mathrm{v} / \mathrm{v}$ ): fractions I-VI.

Fraction III ( 523 g ) was further fractioned by using a silica gel column with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-acetone as the mobile phase (100:0 $\rightarrow 10: 1$ ), yielding five subfractions (Fr. III-1 $\rightarrow$ Fr. III-5). Fr. III-3 ( 37 g ) was further subjected to an $\mathrm{Rp}-18$ column ( $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}=35 \% \rightarrow 100 \%, \mathrm{v} / \mathrm{v}$ ) to afford eight subfractions. Furthermore, Fr. III-3-1 ( 304 mg ) was separated by using a silica gel column (petroleum ether-ethyl acetate) to obtain five parts (1a-1e). Then, Fr. III-3-1a ( 42 mg ) was purified by semipreparative HPLC $\left(\mathrm{CH}_{3} \mathrm{CN}-\mathrm{H}_{2} \mathrm{O}=41 \%\right.$, v/v, flow rate: 3 mL $\mathrm{min}^{-1}$ ) to yield compound $9\left(11.8 \mathrm{mg}, t_{\mathrm{R}}=36.5 \mathrm{~min}\right)$. Fr. III-31 b ( 54 mg ) was subjected to semi-preparative HPLC $\left(\mathrm{CH}_{3} \mathrm{CN}-\right.$ $\mathrm{H}_{2} \mathrm{O}=36 \%$, v/v, flow rate: $3 \mathrm{~mL} \mathrm{~min}{ }^{-1}$ ) to obtain compound 7 $\left(7.2 \mathrm{mg}, t_{\mathrm{R}}=32.6 \mathrm{~min}\right)$. Similarly, compounds $6\left(5.7 \mathrm{mg}, t_{\mathrm{R}}=\right.$ $23.9 \mathrm{~min})$ and $4\left(7.8 \mathrm{mg}, t_{\mathrm{R}}=33.3 \mathrm{~min}\right)$ were respectively obtained from Fr. III-3-1c ( 93 mg ) and Fr. III-3-1d ( 35 mg ) by semi-preparative HPLC $\left(\mathrm{CH}_{3} \mathrm{CN}-\mathrm{H}_{2} \mathrm{O}=55 \%\right.$ and $53 \%$, v/v, flow rate: $3 \mathrm{~mL} \mathrm{~min}{ }^{-1}$ ). Fr. III-3-2 ( 502 mg ) was further subjected to LH-20 (MeOH) to afford six subfractions (2a-2f), of which Fr. III-3-2c was purified to yield compound $10\left(8.2 \mathrm{mg}, t_{\mathrm{R}}=\right.$ $\left.43.8 \mathrm{~min}, t_{\mathrm{R}}=43.8 \mathrm{~min}\right)$ by semi-preparative $\mathrm{HPLC}\left(\mathrm{CH}_{3} \mathrm{CN}-\right.$ $\mathrm{H}_{2} \mathrm{O}=40 \%$, v/v, flow rate: $3 \mathrm{~mL} \min ^{-1}$ ). Additionally, Fr. III-32 d was subjected to semi-preparative $\operatorname{HPLC}\left(\mathrm{CH}_{3} \mathrm{CN}-\mathrm{H}_{2} \mathrm{O}=\right.$ $40 \%, \mathrm{v} / \mathrm{v}$, flow rate: $3 \mathrm{~mL} \mathrm{~min}{ }^{-1}$ ) to obtain compound 1 $\left(5.8 \mathrm{mg}, t_{\mathrm{R}}=34.5 \mathrm{~min}\right)$. Compound $5\left(6.1 \mathrm{mg}, t_{\mathrm{R}}=43.8 \mathrm{~min}\right)$ was isolated from Fr. III-3-2e by semi-preparative HPLC $\left(\mathrm{CH}_{3} \mathrm{CN}-\mathrm{H}_{2} \mathrm{O}=40 \%\right.$, v/v, flow rate: $\left.3 \mathrm{~mL} \mathrm{~min}{ }^{-1}\right)$.

Fr. III-4 $(35 \mathrm{~g})$ was subjected to a silica gel column $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}{ }^{-}\right.$ $\mathrm{MeOH}=80: 1,50: 1,20: 1,10: 1,5: 1, \mathrm{v} / \mathrm{v}$ ) to obtain five fractions: Fr. III-4a-Fr. III-4e. Fr. III-4b was subjected to LH-20 ( MeOH ) to obtain five subfractions ( $4 \mathrm{~b}-1-4 \mathrm{~b}-5$ ). Fr. III-4b-3 was purified by semi-preparative $\operatorname{HPLC}\left(\mathrm{CH}_{3} \mathrm{CN}-\mathrm{H}_{2} \mathrm{O}=55 \%\right.$, v/v, flow rate: $3 \mathrm{~mL} \mathrm{~min}{ }^{-1}$ ) to obtain compound $2\left(5.4 \mathrm{mg}, t_{\mathrm{R}}=\right.$ 33.8 min ). Similarly, $4 \mathrm{~b}-4$ and $4 \mathrm{~b}-5$ were respectively subjected to semi-preparative HPLC $\left(\mathrm{CH}_{3} \mathrm{CN}-\mathrm{H}_{2} \mathrm{O}=42 \%\right.$, $\mathrm{v} / \mathrm{v}$, flow rate: $3 \mathrm{~mL} \mathrm{~min}{ }^{-1}$ ) to yield compounds $8\left(31.7 \mathrm{mg}, t_{\mathrm{R}}=36.5 \mathrm{~min}\right)$ and $3\left(13.2 \mathrm{mg}, t_{\mathrm{R}}=49.3 \mathrm{~min}\right)$.

Ganolearin A (1). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{24} 22.75$ (c 0.16, $\mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\max }(\log \varepsilon): 201$ (3.72), 234 (3.88), and 297 (2.87); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S1; $\dagger$ HRMS (ESI-TOF) $m / z: 401.2322[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{24} \mathrm{H}_{33} \mathrm{O}_{5}$, 401.2323).

Ganolearin B (2). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{20}+51.13$ ( $c$ $0.13, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon): 277$ (2.09), 195 (2.03); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S1; $\dagger$ HRMS (ESI-TOF) $m / z$ : $469.2564[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{26} \mathrm{H}_{38} \mathrm{O}_{6} \mathrm{Na}, 469.2561$ ).
Ganolearin C (3). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{24}-53.39$ (c 0.23, $\mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\text {max }}(\log \varepsilon): 202$ (3.68), and 273 (3.80); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S $1 ; \dagger$ HRMS (ESI-TOF) $\mathrm{m} / \mathrm{z}$ : $457.2602[\mathrm{M}-\mathrm{H}]^{-}$(calcd for $\mathrm{C}_{27} \mathrm{H}_{37} \mathrm{O}_{6}, 457.2596$ ).

Ganolearin D (4). White powder ( MeOH ); $[\alpha]_{\mathrm{D}}^{24} 12.67$ (c 0.18 , $\mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\text {max }}(\log \varepsilon)$ : 202 (3.25), and 256 (3.33); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S1; $\dagger$ HRMS (ESI-TOF) $\mathrm{m} / \mathrm{z}$ : $471.2143[\mathrm{M}+\mathrm{K}]^{+}$(calcd for $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{6} \mathrm{~K}, 471.2143$ ).

Ganolearate E (5). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{24} 97.07$ (c 0.13, $\mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\max }(\log \varepsilon): 202$ (3.72), and 257 (3.77); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S1; $\dagger$ HRMS (ESI-TOF) $m / z$ : $490.3162\left[\mathrm{M}+\mathrm{NH}_{4}\right]^{+}$(calcd for $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{6} \mathrm{NH}_{4}, 490.3163$ ).

Ganolearate $\mathbf{F}$ (6). White powder ( MeOH ); $[\alpha]_{\mathrm{D}}^{23} 7.61$ (c 0.16, $\mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\max }(\log \varepsilon): 201$ (3.72), 234 (3.88), and 297 (2.87); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S1; $\dagger$ HRMS (ESI-TOF) $m / z: 497.2511[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{O}_{7} \mathrm{Na}$, 497.2510).

Ganolearate G (7). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{23}+17.74$ (c $0.10, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\max }(\log \varepsilon): 258$ (2.98), 201 (2.77); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S1; $\dagger$ HRMS (ESI-TOF) $\mathrm{m} /$ $z: 513.2821[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{O}_{7} \mathrm{Na}, 513.2823$ ).

Ganolearic aicd H (8). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{23.9} 20.56$ (c $0.25, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\max }(\log \varepsilon): 203$ (3.58), 254 (3.58), and 270 (3.57); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S2; $\dagger$ HRMS (ESI-TOF) m/z: $473.2550[\mathrm{M} \mathrm{-} \mathrm{H}]^{-}$(calcd for $\mathrm{C}_{27} \mathrm{H}_{37} \mathrm{O}_{7}$, 473.2545).

Ganolearic acid I (9). White powder ( MeOH ); $[\alpha]_{\mathrm{D}}^{23} 8.95$ (c 0.05, MeOH); UV (MeOH); $\lambda_{\max }(\log \varepsilon): 260$ (2.57), 202 (2.78); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S2; $\dagger$ HRMS (ESI-TOF) $m / z: 445.2586[\mathrm{M}+\mathrm{H}]^{+}$(calcd for $\mathrm{C}_{26} \mathrm{H}_{37} \mathrm{O}_{6}$, 445.2585).

Ganolearate J (10). White powder (MeOH); $[\alpha]_{\mathrm{D}}^{24} 14.59$ (c $0.17, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) ; \lambda_{\max }(\log \varepsilon): 202$ (3.29), and 260 (3.54); ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data: see Table S2; $\dagger$ HRMS (ESI-TOF) $m / z: 511.2673[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{7} \mathrm{Na}$, 511.2666).

The crystallographic data (excluding structure factor tables) for the reported structures have been deposited at the Cambridge Crystallographic Data Center (CCDC) as supplementary publication no. CCDC 2258231 for 2 and CCDC 2201713 for 4, and CCDC 2258230 for 5 . Copies of the data can be obtained free of charge from the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax: + 44(0)1223 336 033; e-mail: deposit@ccdc.cam.ac.uk). $\ddagger$
$\ddagger$ Crystal data for 2: $\mathrm{C}_{26} \mathrm{H}_{38} \mathrm{O}_{6}, M=446.56, a=10.8614(3) \AA, b=36.7053(9) \AA, c=$ 6.0922(2) $\AA, \alpha=90^{\circ}, \beta=90^{\circ}, \gamma=90^{\circ}, V=2428.78(12) \AA^{3}, T=150$.(2) K, space group $P 21212, Z=4, \mu(\mathrm{Cu} K \alpha)=0.689 \mathrm{~mm}^{-1}, 15579$ reflections measured, 4402 independent reflections ( $R_{\text {int }}=0.0887$ ). The final $R_{1}$ value was $0.0335(I>2 \sigma(I))$. The final $\mathrm{w} R\left(F^{2}\right)$ value was $0.0856(I>2 \sigma(I))$. The final $R_{1}$ value was 0.0597 (all data). The final $\mathrm{w} R\left(F^{2}\right)$ value was 0.0960 (all data). The goodness of fit on $F^{2}$ was 1.099. Flack parameter $=-0.05(6)$.

Crystal data for 4: $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{6}, M=432.54, a=7.7089(3) \AA, b=8.1588(3) \AA, c=$ $36.7130(13) \AA, \alpha=90^{\circ}, \beta=90^{\circ}, \gamma=90^{\circ}, V=2309.08(15) \AA^{3}, T=150$.(2) K, space group $P 212121, Z=4, \mu(\mathrm{Cu} \mathrm{K} \mathrm{\alpha})=0.709 \mathrm{~mm}^{-1}, 16444$ reflections measured, 4355 independent reflections ( $R_{\text {int }}=0.0890$ ). The final $R_{1}$ value was $0.0449(I>2 \sigma(I))$. The final $\mathrm{w} R\left(F^{2}\right)$ value was $0.1142(I>2 \sigma(I))$. The final $R_{1}$ value was 0.0600 (all data). The final $\mathrm{w} R\left(F^{2}\right)$ value was 0.1190 (all data). The goodness of fit on $F^{2}$ was 1.146. Flack parameter $=0.07(8)$.

Crystal data for 5: $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{6}, M=472.60, a=8.1802(3) \AA, b=13.5202(4) \AA, c=$ $11.2870(4) \AA, \alpha=90^{\circ}, \beta=92.6960(10)^{\circ}, \gamma=90^{\circ}, V=1246.94(7) \AA^{3}, T=150 .(2) \mathrm{K}$, space group P1211, $Z=2, \mu(\mathrm{CuK} \alpha)=0.701 \mathrm{~mm}^{-1}, 22359$ reflections measured, 4854 independent reflections ( $R_{\mathrm{int}}=0.0348$ ). The final $R_{1}$ value was 0.0311 ( $I>$ $2 \sigma(I))$. The final $\mathrm{w} R\left(F^{2}\right)$ value was $0.1015(I>2 \sigma(I))$. The final $R_{1}$ value was 0.0312 (all data). The final $\mathrm{w} R\left(F^{2}\right)$ value was 0.1017 (all data). The goodness of fit on $F^{2}$ was 0.978 . Flack parameter $=-0.09(3)$.

## ECD calculation method for 3

The theoretical calculations of compound 3 were performed using Gaussian 16. ${ }^{31}$ Conformational analysis was carried out. The optimized conformation geometries and thermodynamic parameters of all conformations are provided. The conformers were optimized at the B3LYP/6-311G ( $\mathrm{d}, \mathrm{p}$ ) level. The theoretical calculation of ECD was performed using time-dependent density functional theory (TDDFT) at the B3LYP/6-311G(d,p) level in MeOH with the PCM model. The ECD spectra of compound 3 were obtained by weighing the Boltzmann distribution rate of each geometric conformation.

## Cell culture and treatment

The SH-SY5Y mCherry-GFP-LC3 and SH-SY5Y MAPT cells generated in our previous study were used. ${ }^{6}$ These cells were maintained in Dulbecco's modified Eagle's medium (DMEM) supplemented with $10 \%$ fetal bovine serum (FBS), $1 \times$ MEM nonessential amino acid solution (Gibco, 11140050), $100 \mathrm{U} \mathrm{ml}^{-1}$ penicillin and $100 \mathrm{mg} \mathrm{ml}^{-1}$ streptomycin at $37^{\circ} \mathrm{C}$ in a humidified atmosphere incubator with $5 \% \mathrm{CO}_{2}$ and $95 \%$ humidity, as described in our previous studies. ${ }^{6,32}$ Cells were seeded in prewarmed growth medium in 6 -well plates or 12 -well plates. Rapamycin (InvivoGen, trl-rap) or bafilomycin A1 (tlrl-baf) was used as a positive control. Drugs were applied directly to the culture medium for treatment, and cells were harvested at 24 h after treatment for further analysis.

## Cell counting kit-8 (CCK-8) cell viability assay

For cell viability assays, the CCK-8 kit was used as described by the manufacturer (Beyotime, C0038). Briefly, the SH-SY5Y mCherry-GFP-LC3 cells were plated at $4 \times 10^{3}$ cells per well in 96 -well culture plates. Twenty-four hours after treatment with compounds 1-10, the CCK-8 reagent was added into each well. Cell viability was evaluated by gauging the optical density at 450 nm .

## Flow cytometry analysis

SH-SY5Y mCherry-GFP-LC3 cells, that is, the SH-SY5Y cells stably expressing a triple fusion protein (red fluorescent protein (mCherry), green fluorescent protein (GFP) and the autophagosome marker LC3), ${ }^{33,34}$ which can directly reflect the strength of autophagic flux, were used in detecting the bioactivities of compounds 1-10. These cells show yellow fluorescence due to the co-expression of red mCherry and green GFP in the absence of autophagy. When the autophagy process goes well, autophagosomes and lysosomes fuse to form autolysosomes, and the acidic lysosomal environment quenches the fluorescence of acid-sensitive GFP, while mCherry is not affected, and then the autolysosomes show red fluorescence. Therefore, red fluorescence in the cells can indicate the formation of autolysosomes. ${ }^{34-36}$ A higher red fluorescence and a lower green fluorescence indicate a smoother flux from autophagosomes to autolysosomes. SH-SY5Y mCherry-GFP-LC3 cells were cultured in DMEM supplemented with $10 \%$ fetal bovine serum (Gibco-BRL, 10099-141) at $37^{\circ} \mathrm{C}$ in an incubator with
$5 \% \mathrm{CO}_{2}$ and $95 \%$ humidity. These cells were cultured in 12-well plates for 24 h , and the compound was added directly to the culture medium ( $5 \mu \mathrm{M}$ and $20 \mu \mathrm{M}$ ). Twenty-four hours after the treatment, the cells were harvested and fixed using $4 \%$ paraformaldehyde. The fixed cells were then subjected to a flow cytometry test to check the autophagic flux. Data were analyzed using FlowJo software (FLOWJO, LLC).

## Western blot analysis

For western blotting, the method described in our previous studies was used. ${ }^{22,32,37}$ In brief, the SH-SY5Y and SH-SY5Y MAPT cells were lysed with RIPA lysis buffer (Beyotime, P0013). The protein concentration was determined using the BCA protein assay kit (Beyotime, P0012). A total of $20 \mu \mathrm{~g}$ of protein was separated by $12 \%$ or $8 \%$ ( $\mathrm{vol} / \mathrm{vol}$ ) SDS-polyacrylamide gel and electrophoretically transferred onto a polyvinylidene difluoride membrane (Bio-Rad, L1620177 Rev D). The membrane was soaked with $5 \%$ (wt/vol.) skim milk for 2 h at room temperature. The membrane was incubated with primary antibodies (ACTB [Beijing Zhong Shan-Golden Bridge Biological Technology Co., Ltd, TA-09], GAPDH, glyceralde-hyde-3-phosphate dehydrogenase [Proteintech, 60004-1-Ig], LC3 [Proteintech, 14600-1-AP], phospho-AMPK $\beta 1$ (ser108) (E8N3N) [Cell Signaling Technology, 23021], AMPK (D63G4) [Cell Signaling Technology, 5832], phospho-AMPK $\alpha$ (thr172) (D4D6D) [Cell Signaling Technology, 50081], AMPK $\beta 1 / 2$ (57C12) [Cell Signaling Technology, 4150], mTOR (7C10) [Cell Signaling Technology, 2983], phospho-mTOR (ser2448) (D9C2) [Cell Signaling Technology, 5536], phospho-mTOR (ser2481) [Cell Signaling Technology, 2974], phospho-ULK1 (ser757) (D7O6U) [Cell Signaling Technology, 14202], phospho-ULK1 (ser317) [Cell Signaling Technology, 37762], ULK1 (D8H5) [Cell Signaling Technology, 8054], phospho-ULK1 (ser555) [Cell Signaling Technology, 5869], phospho-ULK1 (ser777) [SigmaAldrich, ABC213], SQSTM1, sequestosome 1 [Elabscience, EAP3350] and Tau (D1M9X) [Cell Signaling Technology, 46687S]) overnight at $4^{\circ} \mathrm{C}$. After three washes with TBST (Trisbuffered saline [Cell Signaling Technology, 9997] with Tween 20 [0.1\%; Sigma, P1379]), each lasting 5 min , the membranes were incubated for 1 h with the peroxidase-conjugated antimouse (474-1806) or antirabbit (474-1516) IgG (1:5000; KPL) at room temperature. The epitope was visualized using an ECL western blot detection kit (Millipore, WBKLS0500). ImageJ software (National Institutes of Health, Bethesda, Maryland, USA) was used for densitometric analysis. GAPDH or ACTB was used as a loading control for the densitometric analysis of the target protein.

## Confocal laser scanning assay

The SH-SY5Y mCherry-GFP-LC3 cells were cultured overnight in DMEM supplemented with $10 \%$ FBS, $1 \times$ MEM nonessential amino acid solution (Gibco, 11140050), $100 \mathrm{U} \mathrm{ml}^{-1}$ penicillin and $100 \mathrm{mg} \mathrm{ml}^{-1}$ streptomycin in a glass-bottom cell culture dish (NEST, 801001). For evaluating tandem fluorescent LC3 puncta, 24 h after treatment with rapamycin and compounds $\mathbf{1}, \mathbf{2}, \mathbf{3}$ or $\mathbf{4}$, the cells were washed three times with PBS and
fixed with $4 \%$ paraformaldehyde. The fixed intact cells were observed using a FluoView 1000 confocal microscope (Olympus, America).

## Statistics

Statistical analysis was performed using GraphPad Prism 8 software. The one-way ANOVA (analysis of variance) was performed using Tukey's post hoc test for comparison between the treated group and the control group, and the values were expressed as mean $\pm$ standard deviation (SD). A $P$-value $<0.05$ was considered statistically significant throughout the study.

## Molecular docking

The crystal structure of the AMPK complex (PDB code: 4CFH) was used for preparing new docking templates through the Swiss-model server. The best configuration of small molecules was refined through energy minimization and molecular docking was performed using Autodock Vina with a center box: $x=-14.105, y=38.735, z=-12.962$ and the dimensions: $72 \times 82 \times 64 \AA$. The docking results were analyzed and shown using Discovery Studio® Visualizer (BIOVIA, San Diego, USA) and PyMOL software (Schrödinger, LLC: NY, USA). ${ }^{5}$

## Author contributions

M. H. Q. and X. R. P. designed this study. X. R. P. supervised the project. Y. L. and D. S. F. performed the isolation and structural analysis. X. R. P. performed the ECD calculation. X. R. P. and R. C. L. wrote the paper. R. C. L. performed cell assays and statistical analysis. Y. G. Y. and M. H. Q. revised the whole manuscript. All authors contributed to the discussion and interpretation of the results.

## Conflicts of interest

There are no conflicts to declare.

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## Novel A-seco-nortriterpenoids from Ganoderma cochlear inhibiting

## Tau pathology by activating AMPK-ULK1-mediated autophagy

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## 1. Structural elucidation of compounds 5-10

The molecular formula of ganolearate $\mathrm{E}(\mathbf{5})($ (Figure $\mathbf{S 1})$ was determined to be $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{6}$ on the basis of HRESIMS ion peak $m / z: 490.3162\left[\mathrm{M}+\mathrm{NH}_{4}\right]^{+}$(calcd. 490.3163). Its 1D NMR spectroscopic data (Table S1) showed similarity with those of cochlearic acid A with the only difference in the presence of an additional methoxyl at $\mathrm{C}-3$, which was confirmed by the HMBC correlations (Figure S2) of OMe ( $\delta_{\mathrm{H}} 3.63$, s) with $\mathrm{C}-3\left(\delta_{\mathrm{C}} 174.1\right)$, of $\mathrm{H}_{2}-1$ and $\mathrm{H}_{2}-2$ with $\mathrm{C}-3$, and of $\mathrm{H}_{3}-19$ with C-1, C-5, and C-10. The further X-ray crystallographic analysis (Figure S3) proved that the absolute configuration of $\mathbf{5}$ was $5 S, 7 S, 10 S, 13 R, 14 R, 17 S, 20 S$. Thus, the structure of compound 5 was determined.

Comparison of molecular weight and 1D NMR spectroscopic data (Table S1) between 6 and $\mathbf{5}$ showed that compound $\mathbf{6}$ had a similar structure with $\mathbf{5}$, except for the absence of the terminal double bond at C-4 and C-28 and the presence of an additional ketone carbonyl ( $\delta_{\mathrm{C}} 214.0$ ). The HMBC correlations of $\mathrm{H}_{3}-29$ with the ketone carbonyl and $\mathrm{C}-5\left(\delta_{\mathrm{C}} 52.5\right)$, and of $\mathrm{H}_{3}-19$ with $\mathrm{C}-1$, C-5, and C-9 ( $\delta_{\mathrm{C}} 137.3$ ) confirmed that C-4 was the ketone carbonyl. Subsequently, the ROESY correlation of $\mathrm{H}-7 / \mathrm{H}_{3}-30$ proved that $7-\mathrm{OH}$ was $\beta$-oriented. Finally, the structure of 6 (ganolearate F) was established.

Ganolearate $\mathrm{G}(7)$ had a molecular formula of $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{O}_{7}$ determined by the HRESIMS $\mathrm{m} / \mathrm{z}$ : $513.2821[\mathrm{M}+\mathrm{Na}]^{+}$(calcd. 513.2823). Meanwhile, compound 7 showed similar 1D NMR spectra with those of compound 6. However, an additional methoxyl group ( $\delta_{\mathrm{H}} 3.67$, $\mathrm{s} ; \delta_{\mathrm{C}} 51.4$ ) was observed in 1D NMR spectra of 7, rather than an oxygenated quaternary carbon in 6. Furthermore, the methoxyl group showed HMBC correlation (Figure S2) with C-24 ( $\delta_{\mathrm{C}} 174.4$ ), simultaneously, $\mathrm{H}_{3}-21$ exhibited a doublet methyl signals at $\delta_{\mathrm{H}} 0.89(\mathrm{~d}, J=6.4 \mathrm{~Hz})$ in the ${ }^{1} \mathrm{H}$ NMR spectrum of 7 . Thus, we speculated that the methoxyl was connected to C-24 and the oxygenated quaternary carbon at C-20 was transformed into a methine. Therefore, the structure of 7 was confirmed.

According to the 1D NMR spectra (Table S2) of 8, the structure of $\mathbf{8}$ resembled that of 7 except for the absence of a methoxyl at C-24 and a ketone carbonyl instead of the hydroxyl at C-7. The HMBC correlations of $\mathrm{H}_{3}-19\left(\delta_{\mathrm{H}} 1.31, \mathrm{~s}\right)$ with C-5 $\left(\delta_{\mathrm{C}} 52.5\right)$ and C-9 $\left(\delta_{\mathrm{C}} 145.6\right)$, of H-5 ( $\delta_{\mathrm{H}} 3.05, \mathrm{t}$, $J=4.5 \mathrm{~Hz})$ with $\mathrm{C}-10\left(\delta_{\mathrm{C}} 39.2\right)$ and C-7 $\left(\delta_{\mathrm{C}} 196.7\right)$, of $\mathrm{H}_{3}-21\left(\delta_{\mathrm{H}} 0.88, \mathrm{~d}, J=6.4 \mathrm{~Hz}\right)$ with C-20 $\left(\delta_{\mathrm{C}}\right.$ 35.6), $\mathrm{C}-17\left(\delta_{\mathrm{C}} 49.0\right)$, and C-22 ( $\delta_{\mathrm{C}} 30.6$ ), and of $\mathrm{H}_{2}-22\left(\delta_{\mathrm{H}} 2.20, \mathrm{~m} ; 2.46, \mathrm{~m}\right)$ with $\mathrm{C}-24\left(\delta_{\mathrm{C}} 178.8\right)$ confirmed above deduction. Thus, the structure of $\mathbf{8}$ was determined and name as ganoclearic acid
H.

The molecular formula of compound 9 was determined to be $\mathrm{C}_{26} \mathrm{H}_{36} \mathrm{O}_{6}$ based on the HRESIMS $m / z: 445.2586[\mathrm{M}+\mathrm{H}]^{+}$(calcd. 445.2585) with 9 degrees of unsaturation. Its ${ }^{1} \mathrm{H}$ NMR spectrum (Table S2) showed that three singlet methyl proton signals at $\delta_{\mathrm{H}} 0.91(\mathrm{~s}), \delta_{\mathrm{H}} 1.25(\mathrm{~s}), \delta_{\mathrm{H}} 1.77(\mathrm{~s})$, one doublet methyl proton signal at $\delta_{\mathrm{H}} 0.88(\mathrm{~d}, J=6.4 \mathrm{~Hz})$, one methoxyl proton signal at $\delta_{\mathrm{H}} 3.63$ (s), one oxymethine proton signal at $\delta_{\mathrm{H}} 4.59(\mathrm{~d}, J=5.2 \mathrm{~Hz})$, and two terminal double bond proton signals at $\delta_{\mathrm{H}} 4.84$ (s) and $\delta_{\mathrm{H}} 4.94$ (s). except for the methoxyl, ${ }^{13} \mathrm{C}$-DEPT NMR spectra of 9 showed 25 carbon resonances, which were assigned as four methyls, eight methylenes (one terminal double bond), four methines (one oxymethine), and nine quaternary carbons (one ester carbonyl, one carboxyl, one ketone, three olefiniccarbons, and one oxygenated carbon). These data suggested that compound $\mathbf{9}$ was a highly degraded lanostane triterpenoid and had a similar $7 / 6 / 5$-tricyclo skeleton with that of cochlate B, which was confirmed by the HMBC correlations (Figure S2) of $\mathrm{H}_{3}-29$ with $\mathrm{C}-4\left(\delta_{\mathrm{C}} 145.8\right), \mathrm{C}-28\left(\delta_{\mathrm{C}} 114.7\right)$, and C-5 ( $\delta_{\mathrm{C}} 54.8$ ), of $\mathrm{H}-5\left(\delta_{\mathrm{H}} 2.75, \mathrm{~m}\right)$ with $\mathrm{C}-1\left(\delta_{\mathrm{C}} 31.8\right), \mathrm{C}-10\left(\delta_{\mathrm{C}}\right.$ 83.5), C-7 ( $\delta_{\mathrm{C}} 75.0$ ), and C-19 ( $\delta_{\mathrm{C}} 36.0$ ), of H-7 ( $\delta_{\mathrm{H}} 4.50, \mathrm{~d}, J=5.2 \mathrm{~Hz}$ ) with C-5, C-9, and C-10, C-14 ( $\delta_{\mathrm{C}} 50.8$ ), and of $\mathrm{H}_{2}-19$ with C-1, C-8 ( $\delta_{\mathrm{C}} 167.9$ ), and C-11 ( $\delta_{\mathrm{C}} 201.2$ ). However, combination the absence of two methylenes and the HMBC correlations (Figure S2) of $\mathrm{H}_{3}-21\left(\delta_{\mathrm{H}} 1.16, \mathrm{~d}, J=\right.$ 6.5 Hz ) with C-17 ( $\delta_{\mathrm{C}} 47.7$ ), C-20 ( $\delta_{\mathrm{C}} 45.1$ ), and C-22 ( $\delta_{\mathrm{C}} 181.7$ ), indicating that $\mathrm{C}-23$ and $\mathrm{C}-24$ were degraded. The ROESY correlations (Figure S2) of $\mathrm{H}-5 / \mathrm{H}_{3}-30 / \mathrm{H}-7$, which demonstrated that H-7 was $\alpha$-oriented. Therefore, the structure of $\mathbf{9}$ was established and named as ganoclearic acid I.

Ganolearate $\mathrm{J}(\mathbf{1 0})$ had a molecular formula of $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{7}$ based on the HRESIMS $m / z: 511.2673$ $[\mathrm{M}+\mathrm{Na}]^{+}$(calcd. 511.2666). Its 1D NMR spectroscopic data (Table S2) were similar with those of cochlate A with the difference in the absence of double bond at C-4 and C-28 and the presence of one oxygenated methylene ( $\delta_{\mathrm{H}} 3.77, \mathrm{~d}, J=12.0 \mathrm{~Hz}, 4.21, \mathrm{~d}, J=12.0 \mathrm{~Hz} ; \delta_{\mathrm{C}} 64.9$ ) and one quaternary carbon containing oxygen ( $\delta_{\mathrm{C}} 86.0$ ). The detailed analysis of HMBC spectrum (Figure S2) of $\mathbf{1 0}$ showed the correlations of $\mathrm{H}_{3}-29\left(\delta_{\mathrm{H}} 1.39, \mathrm{~s}\right)$ and $\mathrm{H}-5\left(\delta_{\mathrm{H}} 2.36, \mathrm{~d}, J=7.6 \mathrm{~Hz}\right)$ with the oxygenated methylene and quaternary carbons, suggesting that C-4 and C-28 were linked to hydroxyl group, respectively. Considering the molecular weight of $\mathbf{1 0}$ and the chemical shift of C-3 ( $\delta_{\mathrm{C}} 175.8$ ) an ester bond between C-4 and C-3 was deduced. Finally, the structure of $\mathbf{1 0}$ was determined.

Table S1. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$-DEPT NMR spectroscopic data of $\mathbf{1 - 7}(600 / 150 \mathrm{MHz}, \delta$ in $p p m, J$ in Hz$)$.


${ }^{a}: \mathrm{CDCl}_{3} ;{ }^{b}: \mathrm{CD}_{3} \mathrm{OD}$

Table S2. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$-DEPT NMR spectroscopic data of $\mathbf{8 - 1 0}(600 / 150 \mathrm{MHz}, \delta$ in $\mathrm{ppm}, J$ in Hz$)$.

| position | $8^{\text {b }}$ |  | $9^{\text {b }}$ |  | $10^{a}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\delta_{\text {H }}$ | $\delta_{\text {C }}$ | $\delta_{\text {H }}$ | $\delta_{\text {C }}$ | $\delta_{\text {H }}$ | $\delta_{\text {C }}$ |
| 1 | 2.07, m; 2.38, m | $33.4 \mathrm{CH}_{2}$ | 1.83, m | $31.8 \mathrm{CH}_{2}$ | 1.86, m | $32.6 \mathrm{CH}_{2}$ |
| 2 | $\begin{aligned} & 2.29, \mathrm{~m} ; \\ & 2.38, \mathrm{~m} \end{aligned}$ | $29.4 \mathrm{CH}_{2}$ | 2.40, m; 2.45, m | $29.5 \mathrm{CH}_{2}$ | 2.47, m; 2.84, m | $27.7 \mathrm{CH}_{2}$ |
| 3 |  | 173.7 C |  | 175.9 C |  | 175.8 C |
| 4 |  | 20.9 C |  | 145.8 C |  | 86.0 C |
| 5 | $3.05, \mathrm{t}$ (4.5) | 52.5 CH | 2.75, m | 54.8 CH | 2.36, d (7.6) | 54.6 CH |
| 6 | 2.52, m; 2.66, m | $36.8 \mathrm{CH}_{2}$ | 2.28, m | $42.1 \mathrm{CH}_{2}$ | $\begin{aligned} & 1.67, \mathrm{~m} \\ & 2.72, \mathrm{~m} \end{aligned}$ | $36.2 \mathrm{CH}_{2}$ |
| 7 |  | 196.7 C | 4.59, d (5.2) | 75.0 CH | 4.64, t (8.9) | 74.4 CH |
| 8 |  | 152.3 C |  | 167.9 C |  | 168.8 C |
| 9 |  | 145.6 C |  | 126.5 C |  | 127.3 C |
| 10 |  | 39.2 C |  | 83.5 C |  | 86.3 C |
| 11 |  | 202.3 C |  | 201.2 C |  | 199.1 C |
| 12 | $2.50, \mathrm{~d}(18.0) ; 2.63, \mathrm{~d}$ <br> (18.0) | $51.1 \mathrm{CH}_{2}$ | $2.44, \mathrm{~d}(17.6) ; 2.78, \mathrm{~d}$ <br> (17.6) | $49.9 \mathrm{CH}_{2}$ | 2.56, d (18.8) | $48.4 \mathrm{CH}_{2}$ |
| 13 |  | 45.2 C |  | 48.8 C |  | 45.8 C |
| 14 |  | 48.4 C |  | 50.8 C |  | 51.0 C |
| 15 | 2.18, m; 2.07, m | $31.7 \mathrm{CH}_{2}$ | 1.27, m; 1.90, m | $30.8 \mathrm{CH}_{2}$ | 1.51, m; 1.91, m | $31.0 \mathrm{CH}_{2}$ |
| 16 | $1.43, \mathrm{~m} ; 2.05, \mathrm{~m}$ | $27.6 \mathrm{CH}_{2}$ | 1.53, m; 2.07, m | $27.7 \mathrm{CH}_{2}$ | 1.51, m; 2.07, m | $26.6 \mathrm{CH}_{2}$ |
| 17 | 1.62 m | 49.0 CH | 2.32, m | 47.7 CH | 1.73, m | 49.6 CH |
| 18 | 0.95, s | $17.7 \mathrm{CH}_{3}$ | 0.91, s | $17.8 \mathrm{CH}_{3}$ | 0.87, s | $17.2 \mathrm{CH}_{3}$ |
| 19 | 1.31 s | $22.0 \mathrm{CH}_{3}$ | 2.06, m; 2.32, m | $36.0 \mathrm{CH}_{2}$ | $3.09, \mathrm{~d}(14.8) ; 3.24, \mathrm{~d}$ <br> (14.8) | $29.6 \mathrm{CH}_{3}$ |
| 20 | $1.45, \mathrm{~m}$ | 35.6 CH | 2.42 m | 45.1 CH | 1.45 , m | 35.7 CH |
| 21 | 0.88, d (6.4) | $17.6 \mathrm{CH}_{3}$ | 1.16, d (6.5) | $17.8 \mathrm{CH}_{3}$ | 0.89, d (6.5) | $17.5 \mathrm{CH}_{3}$ |
| 22 | 2.20, m; 2.46, m | $30.6 \mathrm{CH}_{2}$ |  | 181.7 C | 2.24, m; 2.48, m | $30.7 \mathrm{CH}_{2}$ |
| 23 | 1.89, m;2.08, m | $31.1 \mathrm{CH}_{2}$ |  |  | 1.84, m; 1.34, m | $31.1 \mathrm{CH}_{2}$ |
| 24 |  | 178.8 C |  |  |  | 174.4 C |
| 28 |  |  | 4.84, s; 4.94, s | $\begin{gathered} 114.7 \\ \mathrm{CH}_{2} \end{gathered}$ | 3.77, d (12.0); 4.21, d <br> (12.0) | $64.9 \mathrm{CH}_{2}$ |
| 29 | 2.23, s | $32.6 \mathrm{CH}_{3}$ | 1.77, s | $22.4 \mathrm{CH}_{3}$ | $1.39, \mathrm{~s}$ | $24.9 \mathrm{CH}_{3}$ |
| 30 | 1.08 , s | $24.4 \mathrm{CH}_{3}$ | $1.25, \mathrm{~s}$ | $26.5 \mathrm{CH}_{3}$ | 1.11 , s | $25.7 \mathrm{CH}_{3}$ |
| $\mathrm{OCH}_{3}$ | 3.67 , s | $51.8 \mathrm{CH}_{3}$ | 3.63, s | $52.1 \mathrm{CH}_{3}$ | 3.67, s | $51.5 \mathrm{CH}_{3}$ |
| ${ }^{a}$ : |  | $\mathrm{CDCl}_{3} ;$ |  | ${ }^{b}$ : |  | $\mathrm{CD}_{3} \mathrm{O}$ |








9


Figure S1. Structures of compounds 5-10.


5


9




9

Figure S2. Selected $\mathrm{HMBC}(\mathrm{H} \rightarrow \mathrm{C}),{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY $(\mathrm{H}-\mathrm{H})$ correlations of compounds 5, 6, 9, 10, and ROESY ( correlations of compounds 5 and 9


5

Figure S3. X-ray crystallographic structure of compound 5.
2. Autophagy activation of compounds $\mathbf{1} \mathbf{- 1 0}$

In order to test whether compounds $\mathbf{1} \mathbf{- 1 0}$ would affect autophagy, we used the SH-SY5Y mCherry-GFP-LC3 cell line, which contains thetandem monomeric mCherry-GFP-tagged LC3 (mCherry-GFP-LC3) reporter and was reported in our previous study. ${ }^{[1]}$ The mCherry-GFP-LC3 in autolysosomes displayed more stable red mCherry fluorescence in the acidic lysosome while the GFP signal was sensitive to the acidic condition. ${ }^{[2]}$ DMSO (dimethyl sulfoxide) and rapamycin were respectively as blank and positive controls. ${ }^{[3]}$ The flow cytometry analysis results showed that compounds 1-10 significantly increased the autophagic flux (FigureS4A-B). Moreover, all of them displayed no toxicities in SH-SY5Y mCherry-GFP-LC3 cells by using the CCK-8 assay (FigureS5A). Compounds 1-4 with diverse rearranged skeleton were further investigated for their effect on autophagy and anti-ADs. Treatment with compounds 1-4 can increase autophagic flux in the SH-SY5Y mCherry-GFP-LC3 cells, similar as rapamycin (Figure S5B-C). Collectively, these results demonstrated that compounds 1-4 can activate autophagy.


Figure S4.Compounds $\mathbf{1 - 1 0}$ induced autophagy in SH-SY5Y mCherry-GFP-LC3 cells by flow cytometry analysis. (A) Flow cytometry of SH-SY5Y mCherry-GFP-LC3 cells with or without drug treatment. The percentage of 10,000 cells expressing GFP or/and mCherry were counted. The Q1 area represents the proportion of cells with green fluorescence; the Q2 area represents the proportion of cells with yellow fluorescence; the Q 3 area represents the proportion of cells with only red fluorescence; the Q4 area represents the proportion of cells showing no red and green fluorescence. (B) Quantification of the Q3 area in (A) based on 3 independent experiments. Record the proportion of cells that only emit red fluorescence in Q3 area under each treatment.


Figure S5. Increase of autophagic flux by compound 1-10 in SH-SY5Y mCherry-GFP-LC3 cells.
(A) The CCK-8 assay showing the effects of compound $\mathbf{1 - 1 0}$ on cell viability. (B-C) Increased autophagic flux in response to $\mathbf{1 - 4}$ or Rapamycin (Rapa) treatment in SH-SY5Y mCherry-GFP-LC3 cells. (B) 1-4 treatment increased the maturation of autolysosomes as shown by the increased red puncta of mRFP-GFP-LC3 in cells, and this effect was similar to that of Rapamycin. (C) Quantification of LC3 puncta in (B) based on 3 independent experiments. ns, not significant; *, $P<$ $0.05 ;{ }^{* *}, P<0.01 ;^{* * *}, P<0.001 ;{ }^{* * * *}, P<0.0001$; one-way ANOVA with the Tukey's post-hoc test. Bars represent mean $\pm$ SD.


Figure S6. No inhibition effect of compounds 1-4 on mTOR. (A-H) Western blotting assays showing the protein levels of mTOR, pmTOR2448 and pmTOR2481 in the SH-SY5Y MAPT cells treated with or without compounds. Rapamycin (Rapa) as a positive control. (A-H) A representative Western blotting result $(A, C, E, G)$ and quantification of respective protein levels $(B, D, F, H)$ based on 3 independent experiments were presented. Relative protein abundance was normalized to ACTB. ns, not significant; ${ }^{* *}, P<0.01 ;{ }^{* * *}, P<0.001$; one-way ANOVA with the Tukey's posthoc test. Bars represent mean $\pm$ SD.


Figure S7. The molecular docking results of compounds $\mathbf{1 - 4}$ with AMPK (PDB: 4CFH).

## NMR spectra of new compounds $\mathbf{1 - 1 0}$

Figure S8. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 1 .


Figure $\mathrm{S} 9 .{ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 1.


Figure S10. HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 1.


Figure S11. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 1.


Figure $\mathbf{S 1 2 .}{ }^{1} \mathbf{H}-{ }^{\mathbf{1}} \mathbf{H}$ COSY spectrum $\left(600 \mathrm{MH}, \mathrm{CDCl}_{3}\right)$ of compound 1.


Figure S13. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 1.


Figure S14. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 2.



Figure $\mathrm{S} 15 .{ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 2.


Figure S16. HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 2.


Figure S17. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 2.


Figure S18. ${ }^{\mathbf{1}} \mathbf{H}-\mathbf{}^{\mathbf{H}} \mathrm{H}$ COSY spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 2.


Figure S19. ROESY spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 2.


Figure S20. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 3.

## 




Figure S21. ${ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 3.


Figure S22. HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 3.


Figure S23. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 3 .


Figure S24. ${ }^{\mathbf{1}} \mathbf{H}-{ }^{\mathbf{1}} \mathrm{H}$ COSY spectrum ( $\mathbf{6 0 0} \mathbf{M H}, \mathrm{CDCl}_{3}$ ) of compound 3.


Figure S25. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 3.


Figure S26. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 4.


Figure S27. ${ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 4.


Figure S28. HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 4.


Figure S29. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 4.


Figure S30. ${ }^{\mathbf{1}} \mathrm{H}-\mathbf{}^{\mathbf{1}} \mathrm{H}$ COSY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 4.


Figure S31. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 4.


Figure S32. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 5.



Figure S33. ${ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 5.


Figure S34. HSQC NMR spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 5.


Figure S 35 . HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 5.


Figure S36. ${ }^{\mathbf{1}} \mathrm{H}-\mathbf{}^{\mathbf{H}} \mathrm{H}$ COSY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 5.


Figure S37. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 5.


Figure S38. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CD}_{3} \mathrm{OD}$ ) of compound 6.


Figure S39. ${ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CD}_{3} \mathrm{OD}$ ) of compound 6.


Figure S40. HSQC spectrum (600/150 MH, CD $\left.\mathbf{3}_{3} \mathrm{OD}\right)$ of compound 6.


Figure S 41 . HMBC spectrum $\left(600 / 150 \mathrm{MH}, \mathrm{CD}_{3} \mathrm{OD}\right)$ of compound 6.


Figure $\mathrm{S} 42 .{ }^{1} \mathrm{H}-\mathbf{}^{\mathbf{1}} \mathrm{H}$ COSY spectrum $\left(600 \mathrm{MH}, \mathrm{CD}_{3} \mathrm{OD}\right)$ of compound 6.


Figure S43. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 7 .


Figure $\mathrm{S} 44 .{ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 7.


Figure S 45 . HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 7.


Figure S46. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 7.


Figure $\mathrm{S} 47 .{ }^{1} \mathrm{H}-{ }^{\mathbf{1}} \mathrm{H}$ COSY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 7 .


Figure S48. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 7.


Figure $\mathrm{S} 49 .{ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 8.



Figure $\mathrm{S}_{50} .{ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 8.


Figure S 51 . HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 8.


Figure S52. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 8.


Figure $553 .{ }^{1} \mathrm{H}-{ }^{\mathbf{1}} \mathrm{H}$ COSY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 8.


Figure S54. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 8.


Figure S55. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 9.


Figure S56. ${ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 9 .


Figure S57. HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 9.


Figure S58. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 9.


Figure $\mathbf{S 5 9} .{ }^{1} \mathrm{H}-{ }^{\mathbf{1}} \mathrm{H}$ COSY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 9 .


Figure S60. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 9 .


Figure S61. ${ }^{1} \mathrm{H}$ NMR spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 10.


Figure S62. ${ }^{13} \mathrm{C}$ NMR spectrum ( $150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 10 .


Figure S63. HSQC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 10.


Figure S64. HMBC spectrum ( $600 / 150 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 10.


Figure $\mathrm{S} 65 .{ }^{\mathbf{1}} \mathrm{H}-{ }^{\mathbf{1}} \mathrm{H}$ COSY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 10 .


Figure S66. ROESY spectrum ( $600 \mathrm{MH}, \mathrm{CDCl}_{3}$ ) of compound 10.


## HRESIMS spectra of new compounds

Figure S67. HRESIMS spectrum of compound 1.

## Qualitative Analysis Report



Figure S68. HRESIMS spectrum of compound 2.

## Qualitative Analysis Report

| Data Filename | kfc8-13-13.d | Sample Name | kfc8-13-13 |
| :---: | :---: | :---: | :---: |
| Sample Type | Sample | Position | P1-A5 |
| Instrument Name | Instrument 1 | User Name |  |
| Acq Method | s.m | Acquired Time | 10/27/2022 3:38:48 PM |
| IRM Calibration Status | Success | DA Method | PCDL.m |
| Comment |  |  |  |
| Sample Group |  | Info. |  |
| Acquisition SW | 6200 series TOF/6500 seriesQ-TOF B. 05.01 (B5125.2) |  |  |
| Version |  |  |  |

## User Spectra


Peak List

| $\boldsymbol{m} / \boldsymbol{z}$ | $\mathbf{z}$ | Abund | Formula | Ion |
| :--- | ---: | :--- | :--- | :--- |
| 77.038 | 1 | 715.02 |  |  |
| 96.9608 | 1 | 469.21 |  |  |
| 115.054 | 1 | 891.82 |  |  |
| 128.0618 | 1 | 1282.71 |  |  |
| 165.0692 | 1 | 483.91 |  | $(\mathrm{M}+\mathrm{Na})+$ |
| 167.0841 | 1 | 501.32 |  | $(\mathrm{M}+\mathrm{Na})+$ |
| 469.2564 | 1 | 5928.13 | $\mathrm{C} 26 \mathrm{H} 38 \mathrm{O6}$ |  |
| 470.2603 | 1 | 1448.57 | $\mathrm{C} 26 \mathrm{H} 38 \mathrm{O6}$ |  |
| 485.2298 | 1 | 1013.63 |  |  |
| 922.0098 | 1 | 594.93 |  |  |

Formula Calculator Element Limits

| Element | Min | Max |
| :--- | ---: | :---: |
| C | 3 | 60 |
| H | 0 | 150 |
| O | 0 | 30 |



End Of Report -..

Figure S69. HRESIMS spectrum of compound 3.

Qualitative Analysis Report

| Data Filename | kfD2-1-3.d | Sample Name | kfD2-1-3 |
| :--- | :--- | :--- | :--- |
| Sample Type Sample Position | P1-A2 |  |  |
| Instrument Name | Instrument 1 |  | User Name |




[^1]Figure S70. HRESIMS spectrum of compound 4.


Figure S71. HRESIMS spectrum of compound 5.

Qualitative Analysis Report


Figure S72. HRESIMS spectrum of compound 6.

## Qualitative Analysis Report

|  | KFD2-3-1.d |  | Sample Name |
| :--- | :--- | :--- | :--- | KFD2-3-1

User Spectra

Peak List

| $\boldsymbol{m} / \boldsymbol{z}$ | $\mathbf{z}$ | Abund | Formula | Ion |
| :--- | :--- | :--- | :--- | :--- |
| 202.1893 | 1 | 31287.2 |  |  |
| 202.6902 | 1 | 10662.86 |  |  |
| 403.3687 | 1 | 13301.61 |  |  |
| 439.3191 | 1 | 10997.82 |  |  |
| 497.2511 | 1 | 31826.12 | C27 H38 07 | $(\mathrm{M}+\mathrm{Na})+$ |
| 520.3275 | 1 | 10233.24 |  |  |
| 877.6302 | 1 | 32193.59 |  |  |
| 878.6335 | 1 | 18944.55 |  |  |
| 971.5128 | 1 | 36642.16 |  |  |
| 972.5164 | 1 | 21769.71 |  |  |

Formula Calculator Element Limits

| Element | Min | Max |
| :--- | :--- | :---: |
| C | 3 | 60 |
| H | 0 | 120 |
| O | 0 | 30 |

Formula Calculator Results

| Formula | CalculatedMass | CalculatedMz | Mz | Diff. (mDa) | Diff. (ppm) | DBE |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C27 H38 07 | 474.26 | 497.2510 | 497.2511 | -0.10 | -0.20 | 9.00 |

[^2]Figure S73. HRESIMS spectrum of compound 7.

Qualitative Analysis Report


Figure S74. HRESIMS spectrum of compound 8.

Qualitative Analysis Report


## User Spectra




[^3]Figure S75. HRESIMS spectrum of compound 9.


Figure S76. HRESIMS spectrum of compound 10.

## Qualitative Analysis Report



X-ray crystallographic data of compounds 2,4 , and 5


Figure S77. View of a molecule of compound 2 with the atom-labelling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level.

Table S3. Crystal data and structure refinement for compound 2.

| Identification code | global |  |
| :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{26} \mathrm{H}_{38} \mathrm{O}_{6}$ |  |
| Formula weight | 446.56 |  |
| Temperature | 150(2) K |  |
| Wavelength | 1.54178 A |  |
| Crystal system | Orthorhombic |  |
| Space group | P21212 |  |
| Unit cell dimensions | $\mathrm{a}=10.8614(3) \AA$ | $\alpha=90^{\circ}$. |
|  | $\mathrm{b}=36.7053(9) \AA$ | $\beta=90^{\circ}$. |
|  | $\mathrm{c}=6.0922(2) \AA$ | $\gamma=90^{\circ}$. |
| Volume | 2428.78(12) $\AA 3$ |  |
| Z | 4 |  |
| Density (calculated) | $1.221 \mathrm{mg} / \mathrm{m}^{3}$ |  |
| Absorption coefficient | $0.689 \mathrm{~mm}^{-1}$ |  |
| $\mathrm{F}(000)$ | 968 |  |
| Crystal size | $0.250 \times 0.120 \times 0.070 \mathrm{~mm}^{3}$ |  |
| Theta range for data collection | 2.41 to $68.32^{\circ}$. |  |
| Index ranges | $-13<=\mathrm{h}<=12,-44<=\mathrm{k}<=34,-7<=\mathrm{l}<=6$ |  |
| Reflections collected | 15579 |  |
| Independent reflections | $4402[\mathrm{R}(\mathrm{int})=0.0887]$ |  |
| Completeness to theta $=68.32^{\circ}$ | 99.80\% |  |
| Absorption correction | Semi-empirical from equivalents |  |
| Max. and min. transmission | 0.95 and 0.79 |  |
| Refinement method | Full-matrix least-squares on F2 |  |
| Data / restraints / parameters | 4402 / 0 / 296 |  |
| Goodness-of-fit on F2 | 1.099 |  |
| Final R indices [I>2sigma(I)] | $\mathrm{R} 1=0.0335, \mathrm{wR} 2=0.0856$ |  |
| R indices (all data) | $\mathrm{R} 1=0.0597, \mathrm{wR} 2=0.0960$ |  |
| Absolute structure parameter | -0.05(6) |  |



Figure S78. View of a molecule of compound 4 with the atom-labelling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level.

Table S4. Crystal data and structure refinement for compound 4.

| Identification code | global |  |
| :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{6}$ |  |
| Formula weight | 432.54 |  |
| Temperature | 150(2) K |  |
| Wavelength | 1.54178 A |  |
| Crystal system | Orthorhombic |  |
| Space group | P212121 |  |
| Unit cell dimensions | $\mathrm{a}=7.7089(3) \AA$ | $\alpha=90^{\circ}$. |
|  | $\mathrm{b}=8.1588(3) \AA$ | $\beta=90^{\circ}$. |
|  | $\mathrm{c}=36.7130(13) \AA$ | $\gamma=90^{\circ}$. |
| Volume | 2309.08(15) $\AA 3$ |  |
| Z | 4 |  |
| Density (calculated) | $1.244 \mathrm{mg} / \mathrm{m}^{3}$ |  |
| Absorption coefficient | $0.709 \mathrm{~mm}^{-1}$ |  |
| $\mathrm{F}(000)$ | 936 |  |
| Crystal size | $0.680 \times 0.190 \times 0.040 \mathrm{~mm}^{3}$ |  |
| Theta range for data collection | 2.41 to $69.95^{\circ}$. |  |
| Index ranges | $-9<=\mathrm{h}<=9,-7<=\mathrm{k}<=9,-44<=1<=44$ |  |
| Reflections collected | 16444 |  |
| Independent reflections | $4355[\mathrm{R}(\mathrm{int})=0.0890]$ |  |
| Completeness to theta $=69.95^{\circ}$ | 99.90\% |  |
| Absorption correction | Semi-empirical from equivalents |  |
| Max. and min. transmission | 0.97 and 0.78 |  |
| Refinement method | Full-matrix least-squares on F2 |  |
| Data / restraints / parameters | 4355 / 0 / 286 |  |
| Goodness-of-fit on F2 | 1.146 |  |
| Final R indices [ $\mathrm{I}>2$ sigma( I )] | $\mathrm{R} 1=0.0449, \mathrm{wR} 2=0.1142$ |  |
| R indices (all data) | $\mathrm{R} 1=0.0600, \mathrm{wR} 2=0.1190$ |  |
| Absolute structure parameter | 0.07(8) |  |
| Largest diff. peak and hole | 0.482 and -0.454 e.Å-3 |  |



Figure S79. View of a molecule of compound 5 with the atom-labelling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level.

Table S5. Crystal data and structure refinement for compound 5.

| Identification code | global |  |
| :---: | :---: | :---: |
| Empirical formula | $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{6}$ |  |
| Formula weight | 472.6 |  |
| Temperature | 150(2) K |  |
| Wavelength | 1.54178 A |  |
| Crystal system | Monoclinic |  |
| Space group | P 1211 |  |
| Unit cell dimensions | $\mathrm{a}=8.1802(3) \AA$ | $\alpha=90^{\circ}$. |
|  | $\mathrm{b}=13.5202(4) \AA$ | $\beta=90^{\circ}$. |
|  | $\mathrm{c}=11.2870(4) \AA$ | $\gamma=90^{\circ}$. |
| Volume | 1246.94(7) Å 3 |  |
| Z | 2 |  |
| Density (calculated) | $1.259 \mathrm{mg} / \mathrm{m}^{3}$ |  |
| Absorption coefficient | $0.701 \mathrm{~mm}^{-1}$ |  |
| $\mathrm{F}(000)$ | 512 |  |
| Crystal size | $0.540 \times 0.450 \times 0.280 \mathrm{~mm}^{3}$ |  |
| Theta range for data collection | 3.92 to $72.31^{\circ}$. |  |
| Index ranges | $-8<=\mathrm{h}<=10,-16<=\mathrm{k}<=16,-13<=\mathrm{l}<=13$ |  |
| Reflections collected | 22359 |  |
| Independent reflections | 4854 [ $\mathrm{R}(\mathrm{int}$ ) $=0.0348]$ |  |
| Completeness to theta $=69.95^{\circ}$ | 99.00\% |  |
| Absorption correction | Semi-empirical from equivalents |  |
| Max. and min. transmission | 0.83 and 0.69 |  |
| Refinement method | Full-matrix least-squares on F2 |  |
| Data / restraints / parameters | 4854 / 1 / 314 |  |
| Goodness-of-fit on F2 | 0.978 |  |
| Final R indices [I>2sigma(I)] | $\mathrm{R} 1=0.0311, \mathrm{wR} 2=0.1015$ |  |
| R indices (all data) | $\mathrm{R} 1=0.0312, \mathrm{wR} 2=0.1017$ |  |
| Absolute structure parameter | -0.09(3) |  |
| Largest diff. peak and hole | 0.213 and -0.214 e. $\AA$ - 3 |  |

## Calculated ECD data of compound 3



File: compound 2-1mm (195-400) 22090803.dsx
ProBinaryX
Attributes :

- Time Stamp: Thu Sep 08 13:38:16 2022
- File ID : F361B51F-DFB3-4086-8AA1-5D9E05E7A180
- Is CFR Compliant: false
- Original data has not been modified.

Remarks:

- User: CD
- Date: 2022/09/08
- Instrument: 0547
- DetectorType: LAAPD
- DichOS Calibration Correction Curve: 0547/2
- HV (CDDC channel): 0 v
- Time per point: 1 s
- Description: Sample 1
- Concentration: $0.4600 \mathrm{mg} / \mathrm{mL} \mathrm{MeOH}$
- Pathlength: 1 mm
- Temperature: $20^{\circ} \mathrm{C}$

Settings:

- Time-per-point: 1s (25us x 40000)
- SE
- Wavelength: 195nm - 400nm
- Step Size: 1 nm
- Bandwidth: 1nm
$\begin{array}{lllllll}\text { Figure } & \text { S80. The } & \text { CD } & \text { spectrum } & \text { of }\end{array}$


3


3-1


3-4





Figure S81. Six optimized conformers of 3.
Table S6. Conformational analysis of the six optimized conformers of 3 in the gas phase ( $\mathbf{T}=$ 298.15 K)

| Conformer | $\mathrm{E}($ Hartree $)$ | $\mathrm{C}($ Hartree $)$ | $\mathrm{G}(\mathrm{kcal} / \mathrm{mol})$ | $\Delta \mathrm{G}(\mathrm{kcal} / \mathrm{mol})$ | Population |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{3 - 1}$ | -1502.963528 | 0.537406 | -942787.415691 | 0 | $50.38 \%$ |
| $\mathbf{3 - 2}$ | -1502.961835 | 0.536663 | -942786.819782 | 0.595908597 | $18.42 \%$ |
| $\mathbf{3 - 3}$ | -1502.961967 | 0.537037 | -942786.667542 | 0.748148798 | $14.24 \%$ |
| $\mathbf{3 - 4}$ | -1502.961837 | 0.537663 | -942786.193728 | 1.221962773 | $6.40 \%$ |
| $\mathbf{3 - 5}$ | -1502.962021 | 0.537875 | -942786.175875 | 1.239815433 | $6.21 \%$ |
| $\mathbf{3 - 6}$ | -1502.963046 | 0.539234 | -942785.965986 | 1.449704978 | $4.35 \%$ |

Electronic energy obtained at M062X/6-311+G(2d,p) EmpiricalDispersion=GD3 of theory; Thermal correction to Gibbs free energy obtained at B3LYP/6-31G(d) Scale=0.9813 SCRF=(IEFPCM,Solvent=Methanol) EmpiricalDispersion=GD3BJ of theory; Gibbs free energy (E + C); The relative Gibbs free energy; The Boltzmann distribution of each conformer.

Table S7. Atomic coordinates $(\AA)$ of 3-1 obtained at the Cam-B3LYP/6-311+G(d,p) level of theory in the MeOH .

| C | -3.23512 | 0.526145 | -0.56739 | H | 0.119621 | -2.313954 | 0.32708 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C | -2.122056 | 2.896378 | -0.333574 | H | 0.932854 | 1.843407 | 2.580452 |
| C | -1.13449 | 0.773697 | 0.569663 | H | 1.159912 | 2.32409 | 0.903301 |
| C | -1.824834 | -0.010336 | -0.300388 | H | 3.317207 | 1.374559 | 1.078659 |
| C | 0.19691 | 0.375992 | 1.122447 | H | 2.906236 | 0.525473 | 2.556464 |
| C | 1.018129 | -0.352274 | -0.001508 | H | 2.240756 | -1.418372 | 1.42333 |
| C | 0.207584 | -1.568463 | -0.473631 | H | 1.875098 | 1.435588 | -0.975986 |
| C | -1.219404 | -1.201115 | -0.89923 | H | 1.752793 | 0.019954 | -2.016593 |
| C | 1.178004 | 1.466868 | 1.582592 | H | 0.308422 | 0.950923 | -1.62946 |
| C | 2.5539 | 0.744099 | 1.544493 | H | -0.729933 | -1.391108 | 2.084626 |
| C | 2.37444 | -0.581764 | 0.725174 | H | 0.786959 | -0.862865 | 2.855399 |
| C | 1.24866 | 0.574155 | -1.219442 | H | -0.695622 | 0.065459 | 3.074908 |
| C | -0.12209 | -0.524864 | 2.351256 | H | 3.808412 | -0.068877 | -0.792381 |
| O | -1.811074 | -1.911788 | -1.716835 | H | 3.074056 | -3.015872 | -0.36453 |
| C | -1.894902 | 1.997591 | 0.899677 | H | 4.370407 | -2.489437 | -1.446797 |
| C | 3.615907 | -0.918382 | -0.125625 | H | 2.708263 | -2.027755 | -1.78732 |
| C | 3.428958 | -2.179904 | -0.980128 | H | 4.840201 | -0.320261 | 1.575478 |
| C | 4.850734 | -1.078924 | 0.789155 | H | 4.800665 | -2.054169 | 1.290153 |
| C | 6.187484 | -0.969472 | 0.053952 | H | 7.015565 | -1.257482 | 0.715448 |
| C | 6.485584 | 0.434647 | -0.422474 | H | 6.250334 | -1.643829 | -0.806332 |
| O | 7.5108 | 0.453524 | -1.303706 | H | 7.666171 | 1.387841 | -1.542075 |
| O | 5.917714 | 1.449686 | -0.0675 | H | -4.587147 | 1.070781 | -2.180456 |
| C | -3.585291 | 0.643382 | -2.058433 | H | -3.554387 | -0.332164 | -2.546478 |
| C | -4.320083 | -0.278333 | 0.203834 | H | -2.872753 | 1.295602 | -2.573463 |
| C | -3.173286 | 1.904277 | 0.119222 | H | -4.073245 | -0.255477 | 1.270618 |
| C | -2.38695 | 4.33414 | -0.060752 | H | -5.263685 | 0.2677 | 0.081917 |
| C | -1.86563 | 5.315466 | -1.085307 | H | -2.123343 | 6.336742 | -0.798902 |
| O | -2.990992 | 4.69569 | 0.942761 | H | -0.776564 | 5.21837 | -1.174124 |
| C | -4.562293 | -1.732381 | -0.210113 | H | -2.285984 | 5.088097 | -2.072495 |
| C | -3.570128 | -2.718017 | 0.366447 | H | -5.550165 | -2.046474 | 0.15403 |
| O | -3.611285 | -3.887548 | -0.295365 | H | -4.58428 | -1.858993 | -1.294175 |
| O | -2.850457 | -2.523466 | 1.329283 | H | -1.651088 | -4.534083 | -0.035085 |
| C | -2.671157 | -4.884598 | 0.141134 | H | -2.804144 | -5.10518 | 1.202588 |
| H | -4.086743 | 2.334432 | 0.517025 | H | -2.878612 | -5.767613 | -0.462583 |
| H | -1.809333 | 2.499346 | 1.857341 |  |  |  |  |
| H | -1.539323 | 2.667968 | -1.220473 |  |  |  |  |
| H | 0.663282 | -2.079206 | -1.324518 |  |  |  |  |

Table S8. Atomic coordinates $(\AA$ ) of 3-2 obtained at the Cam-B3LYP/6-311+G(d,p) level of theory in the $\mathbf{M e O H}$.

| C | 2.79563 | -0.680457 | 1.390819 | H | -0.971786 | -2.854745 | 0.019287 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C | 2.253769 | 1.860815 | 0.967764 | H | -0.512386 | 1.333837 | -2.353118 |
| C | 0.992294 | -0.055739 | -0.061568 | H | -0.853291 | 1.911142 | -0.726491 |
| C | 1.375258 | -0.941184 | 0.894516 | H | -3.119431 | 1.416268 | -1.174525 |
| C | -0.299256 | -0.200235 | -0.799252 | H | -2.70924 | 0.467029 | -2.589887 |
| C | -1.396287 | -0.716837 | 0.201496 | H | -2.598587 | -1.545554 | -1.387277 |
| C | -0.933739 | -2.064621 | 0.779735 | H | -2.328413 | -0.135438 | 2.0802 |
| C | 0.501252 | -2.030152 | 1.332218 | H | -0.66748 | 0.431175 | 1.933556 |
| C | -0.961535 | 1.0526 | -1.395634 | H | -1.961149 | 1.24348 | 1.047634 |
| C | -2.450815 | 0.631029 | -1.540007 | H | 0.237855 | -2.187098 | -1.617189 |
| C | -2.657523 | -0.683452 | -0.709736 | H | -0.860791 | -1.257846 | -2.654587 |
| C | -1.593906 | 0.268625 | 1.377489 | H | 0.851171 | -0.819779 | -2.537107 |
| C | -0.011174 | -1.183595 | -1.970829 | H | -4.152325 | 0.141132 | 0.596154 |
| O | 0.880276 | -2.912542 | 2.106212 | H | -3.984226 | -2.902256 | 0.252736 |
| C | 2.02811 | 0.980112 | -0.272846 | H | -5.279195 | -2.099291 | 1.151181 |
| C | -4.046809 | -0.74386 | -0.043294 | H | -3.619491 | -1.979677 | 1.719159 |
| C | -4.239964 | -1.998527 | 0.819845 | H | -4.87735 | 0.052342 | -1.894535 |
| C | -5.15073 | -0.67031 | -1.121889 | H | -5.233152 | -1.6464 | -1.616907 |
| C | -6.524701 | -0.274218 | -0.579214 | H | -7.295821 | -0.403862 | -1.3504 |
| C | -6.591822 | 1.171867 | -0.14129 | H | -6.84135 | -0.901106 | 0.260874 |
| O | -7.711346 | 1.422968 | 0.573971 | H | -7.702476 | 2.374237 | 0.795008 |
| O | -5.777991 | 2.038676 | -0.396402 | H | 2.250285 | 0.219608 | 3.305303 |
| C | 2.896829 | -0.575603 | 2.921481 | H | 3.927143 | -0.350532 | 3.219686 |
| C | 3.765093 | -1.793494 | 0.908793 | H | 2.594346 | -1.517018 | 3.386211 |
| C | 3.128508 | 0.65176 | 0.697344 | H | 4.78745 | -1.502312 | 1.174859 |
| C | 2.874484 | 3.188217 | 0.708724 | H | 3.529027 | -2.701568 | 1.471526 |
| C | 2.429324 | 4.322631 | 1.601415 | H | 1.346039 | 4.467469 | 1.507354 |
| O | 3.69805 | 3.346568 | -0.186403 | H | 2.626073 | 4.073882 | 2.651334 |
| C | 3.710207 | -2.156426 | -0.586911 | H | 2.949992 | 5.244066 | 1.334669 |
| C | 4.115498 | -1.061826 | -1.549684 | H | 4.392114 | -2.999647 | -0.754726 |
| O | 5.314822 | -0.535429 | -1.22911 | H | 2.709619 | -2.483114 | -0.87513 |
| O | 3.46957 | -0.699432 | -2.515736 | H | 5.104361 | 1.451901 | -1.812029 |
| C | 5.742739 | 0.589136 | -2.022653 | H | 6.767096 | 0.791887 | -1.711836 |
| H | 4.160319 | 0.87564 | 0.45736 | H | 5.702501 | 0.347191 | -3.086178 |
| H | 2.199825 | 1.43578 | -1.240555 |  |  |  |  |
| H | 1.51023 | 1.812324 | 1.756957 |  |  |  |  |
| H | -1.568952 | -2.407157 | 1.600183 |  |  |  |  |

Table S9. Atomic coordinates $(\AA)$ of 3-3 obtained at the Cam-B3LYP/6-311+G(d,p) level of theory in the MeOH .

| C | -2.830567 | -0.689934 | -1.430543 | H | 0.892765 | -2.84227 | 0.038079 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C | -2.302268 | 1.857478 | -1.056942 | H | 0.475589 | 1.427938 | 2.257021 |
| C | -1.026725 | -0.032334 | 0.008477 | H | 0.837853 | 1.941083 | 0.613658 |
| C | -1.409647 | -0.939055 | -0.9275 | H | 3.094428 | 1.438762 | 1.105227 |
| C | 0.259729 | -0.160121 | 0.75781 | H | 2.658246 | 0.54156 | 2.546426 |
| C | 1.359086 | -0.721254 | -0.214889 | H | 2.541277 | -1.508739 | 1.410107 |
| C | 0.885565 | -2.081511 | -0.75289 | H | 1.959183 | 1.201644 | -1.121502 |
| C | -0.537057 | -2.041249 | -1.335537 | H | 2.30721 | -0.215726 | -2.107291 |
| C | 0.930152 | 1.106682 | 1.314781 | H | 0.655339 | 0.383143 | -1.986073 |
| C | 2.413026 | 0.673549 | 1.488895 | H | 0.801507 | -1.168878 | 2.647426 |
| C | 2.613471 | -0.670252 | 0.704768 | H | -0.894316 | -0.685517 | 2.523304 |
| C | 1.576829 | 0.222927 | -1.421219 | H | -0.326404 | -2.103802 | 1.647033 |
| C | -0.047804 | -1.095721 | 1.963396 | H | 4.126295 | 0.096507 | -0.615448 |
| O | -0.908871 | -2.930915 | -2.104524 | H | 3.924386 | -2.932033 | -0.174169 |
| C | -2.063206 | 1.004648 | 0.199524 | H | 5.236512 | -2.172262 | -1.085105 |
| C | 4.007236 | -0.765801 | 0.051932 | H | 3.58383 | -2.054028 | -1.673538 |
| C | 4.195208 | -2.050011 | -0.767817 | H | 4.829794 | 0.081772 | 1.884029 |
| C | 5.103201 | -0.66705 | 1.136663 | H | 5.17363 | -1.627662 | 1.662826 |
| C | 6.484558 | -0.299443 | 0.592714 | H | 7.248947 | -0.412112 | 1.373189 |
| C | 6.56757 | 1.131867 | 0.111187 | H | 6.801732 | -0.954421 | -0.225498 |
| O | 7.696778 | 1.352557 | -0.598797 | H | 7.697539 | 2.296544 | -0.849313 |
| O | 5.757537 | 2.012038 | 0.330116 | H | -2.275607 | 0.182588 | -3.356375 |
| C | -2.925414 | -0.605223 | -2.963213 | H | -3.954124 | -0.382581 | -3.268373 |
| C | -3.789251 | -1.809396 | -0.942614 | H | -2.622826 | -1.55372 | -3.413265 |
| C | -3.171356 | 0.651513 | -0.754291 | H | -4.810628 | -1.556317 | -1.250216 |
| C | -2.921677 | 3.189853 | -0.819688 | H | -3.508477 | -2.729734 | -1.462069 |
| C | -2.517259 | 4.295465 | -1.766348 | H | -3.039765 | 5.220695 | -1.516836 |
| O | -3.713059 | 3.371713 | 0.09921 | H | -1.433296 | 4.454845 | -1.710918 |
| C | -3.783964 | -2.096906 | 0.568704 | H | -2.742998 | 4.006658 | -2.800061 |
| C | -4.448425 | -1.006365 | 1.377907 | H | -4.359488 | -3.013382 | 0.746868 |
| O | -3.676955 | -0.585561 | 2.398672 | H | -2.771429 | -2.2675 | 0.937886 |
| O | -5.553022 | -0.549193 | 1.146986 | H | -5.166096 | 0.246913 | 3.60922 |
| C | -4.205163 | 0.514614 | 3.164959 | H | -3.464761 | 0.711762 | 3.939163 |
| H | -4.201736 | 0.87365 | -0.504086 | H | -4.330334 | 1.389366 | 2.521302 |
| H | -2.221685 | 1.492172 | 1.153875 |  |  |  |  |
| H | -1.567647 | 1.79206 | -1.853187 |  |  |  |  |
| H | 1.531188 | -2.465897 | -1.545966 |  |  |  |  |

Table S10. Atomic coordinates $(\AA)$ of 3-4 obtained at the Cam-B3LYP/6-311+G(d,p) level of theory in the MeOH .

| C | -3.153679 | 0.39634 | -0.869514 | H | 0.165579 | -2.356128 | 0.316631 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C | -2.25723 | 2.804436 | -0.325539 | H | 0.830543 | 1.911184 | 2.428515 |
| C | -1.158136 | 0.670856 | 0.445269 | H | 1.080915 | 2.31793 | 0.734748 |
| C | -1.761872 | -0.109811 | -0.48915 | H | 3.265316 | 1.451443 | 1.011653 |
| C | 0.179718 | 0.347103 | 1.029811 | H | 2.840126 | 0.650505 | 2.511939 |
| C | 1.049861 | -0.390264 | -0.051639 | H | 2.271354 | -1.360721 | 1.441925 |
| C | 0.288462 | -1.64443 | -0.509325 | H | 1.811142 | -0.068956 | -2.065501 |
| C | -1.118181 | -1.312966 | -1.019991 | H | 0.354146 | 0.862516 | -1.730203 |
| C | 1.110818 | 1.495965 | 1.455701 | H | 1.905356 | 1.377159 | -1.06596 |
| C | 2.509419 | 0.816122 | 1.482791 | H | -0.679563 | -1.409939 | 2.073545 |
| C | 2.394537 | -0.548252 | 0.714271 | H | 0.801346 | -0.770312 | 2.830548 |
| C | 1.289163 | 0.504273 | -1.29344 | H | -0.723667 | 0.097268 | 2.983656 |
| C | -0.115457 | -0.504606 | 2.298342 | H | 3.8517 | -0.045797 | -0.785604 |
| O | -1.68584 | -2.065692 | -1.816335 | H | 2.843399 | -2.075192 | -1.732524 |
| C | -2.023106 | 1.81157 | 0.823414 | H | 3.201572 | -2.997333 | -0.263967 |
| C | 3.668281 | -0.875044 | -0.091571 | H | 4.509399 | -2.468324 | -1.331091 |
| C | 3.545924 | -2.173363 | -0.901354 | H | 4.826084 | -0.170164 | 1.61498 |
| C | 4.881967 | -0.959133 | 0.860964 | H | 4.8481 | -1.914488 | 1.400225 |
| C | 6.234886 | -0.836928 | 0.158195 | H | 7.052452 | -1.071784 | 0.852909 |
| C | 6.503581 | 0.555165 | -0.36826 | H | 6.342683 | -1.54323 | -0.671542 |
| O | 7.560356 | 0.572171 | -1.211447 | H | 7.69287 | 1.500197 | -1.485466 |
| O | 5.889047 | 1.564056 | -0.079616 | H | -2.551874 | 1.402438 | -2.7229 |
| C | -3.288561 | 0.670662 | -2.379223 | H | -4.287406 | 1.061137 | -2.605877 |
| C | -4.291388 | -0.578107 | -0.46373 | H | -3.139199 | -0.25503 | -2.940208 |
| C | -3.248552 | 1.690496 | -0.035022 | H | -5.227998 | -0.008896 | -0.472972 |
| C | -2.660002 | 4.18243 | 0.065526 | H | -4.385492 | -1.350624 | -1.229766 |
| C | -2.175078 | 5.300231 | -0.828482 | H | -1.078728 | 5.300353 | -0.864459 |
| O | -3.345229 | 4.390386 | 1.060187 | H | -2.526118 | 5.139874 | -1.85516 |
| C | -4.133573 | -1.265266 | 0.908692 | H | -2.532815 | 6.263637 | -0.460703 |
| C | -3.354691 | -2.56227 | 0.841358 | H | -3.645224 | -0.615372 | 1.637995 |
| O | -3.914723 | -3.421221 | -0.028025 | H | -5.128533 | -1.513837 | 1.296576 |
| O | -2.359121 | -2.834459 | 1.488555 | H | -2.223254 | -4.406064 | -0.71987 |
| C | -3.186715 | -4.64079 | -0.262524 | H | -3.035051 | -5.184615 | 0.672293 |
| H | -4.214811 | 2.010564 | 0.341484 | H | -3.804209 | -5.221139 | -0.947476 |
| H | -2.032011 | 2.219882 | 1.82853 |  |  |  |  |
| H | -1.612736 | 2.715253 | -1.193926 |  |  |  |  |
| H | 0.796317 | -2.181827 | -1.312937 |  |  |  |  |

Table S11. Atomic coordinates $(\AA)$ of 3-5 obtained at the Cam-B3LYP/6-311+G(d,p) level of theory in the MeOH.

| C | -2.741717 | -0.666045 | -1.438733 | H | 1.083615 | -2.745604 | -0.062543 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C | -2.310669 | 1.87572 | -0.90264 | H | 0.395446 | 1.312848 | 2.470357 |
| C | -0.999385 | -0.030994 | 0.081966 | H | 0.737608 | 1.979672 | 0.878608 |
| C | -1.324888 | -0.89213 | -0.916478 | H | 3.006452 | 1.569902 | 1.327184 |
| C | 0.279491 | -0.155577 | 0.845111 | H | 2.636284 | 0.557557 | 2.710401 |
| C | 1.41992 | -0.585294 | -0.149408 | H | 2.612853 | -1.422983 | 1.440584 |
| C | 1.025244 | -1.926744 | -0.790438 | H | 1.91184 | 1.431506 | -0.904371 |
| C | -0.399133 | -1.929626 | -1.371615 | H | 2.377547 | 0.112321 | -1.974998 |
| C | 0.873213 | 1.096836 | 1.509957 | H | 0.687019 | 0.591959 | -1.859498 |
| C | 2.377934 | 0.738412 | 1.663529 | H | -0.19585 | -2.195056 | 1.567657 |
| C | 2.656637 | -0.536561 | 0.794611 | H | 0.840718 | -1.267746 | 2.668805 |
| C | 1.60352 | 0.453107 | -1.28106 | H | -0.883948 | -0.892243 | 2.527304 |
| C | 0.004806 | -1.198242 | 1.967806 | H | 4.159559 | 0.394443 | -0.429606 |
| O | -0.725929 | -2.7986 | -2.183496 | H | 5.365385 | -1.69536 | -1.130387 |
| C | -2.079717 | 0.955225 | 0.307519 | H | 3.675929 | -1.738039 | -1.603678 |
| C | 4.068473 | -0.52065 | 0.172456 | H | 4.212697 | -2.665695 | -0.195065 |
| C | 4.343733 | -1.722454 | -0.739982 | H | 4.888145 | 0.404642 | 1.956324 |
| C | 5.116664 | -0.44118 | 1.302619 | H | 5.051538 | -1.346039 | 1.920661 |
| C | 6.575923 | -0.292119 | 0.825455 | H | 7.217864 | -0.115102 | 1.695808 |
| C | 6.737985 | 0.861437 | -0.133408 | H | 6.933734 | -1.191238 | 0.321529 |
| O | 6.443674 | 2.047028 | 0.449834 | H | 6.540002 | 2.735044 | -0.236373 |
| O | 7.063249 | 0.775866 | -1.301526 | H | -3.839492 | -0.30849 | -3.282996 |
| C | -2.808691 | -0.504198 | -2.966383 | H | -2.456885 | -1.41385 | -3.458944 |
| C | -3.678603 | -1.833706 | -1.026834 | H | -2.185505 | 0.330404 | -3.302092 |
| C | -3.143063 | 0.623687 | -0.702258 | H | -4.705085 | -1.569566 | -1.304961 |
| C | -2.991121 | 3.166097 | -0.608185 | H | -3.395517 | -2.707482 | -1.621169 |
| C | -2.577908 | 4.350426 | -1.449973 | H | -2.744731 | 4.132727 | -2.511908 |
| O | -3.837808 | 3.256555 | 0.274664 | H | -3.142208 | 5.238674 | -1.160352 |
| C | -3.643038 | -2.257679 | 0.453223 | H | -1.503682 | 4.537042 | -1.329234 |
| C | -4.109555 | -1.220394 | 1.450908 | H | -4.296604 | -3.13141 | 0.569828 |
| O | -5.320633 | -0.725182 | 1.12529 | H | -2.637562 | -2.559784 | 0.751013 |
| O | -3.498354 | -0.875867 | 2.445665 | H | -6.831169 | 0.527228 | 1.628665 |
| C | -5.807599 | 0.347491 | 1.955771 | H | -5.783328 | 0.060354 | 3.008489 |
| H | -4.188245 | 0.79773 | -0.479434 | H | -5.196455 | 1.24079 | 1.798604 |
| H | -2.292345 | 1.364392 | 1.287827 |  |  |  |  |
| H | -1.547505 | 1.887842 | -1.674279 |  |  |  |  |
| H | 1.687545 | -2.208879 | -1.612406 |  |  |  |  |

Table S12. Atomic coordinates $(\AA)$ of 3-6 obtained at the Cam-B3LYP/6-311+G(d,p) level of theory in the MeOH .

| C | 2.632342 | 0.016763 | 1.567426 | H | -1.761599 | -1.181504 | 1.800101 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C | 2.698618 | 1.835883 | -0.334578 | H | -0.71321 | 0.401955 | -2.644578 |
| C | 0.806003 | 0.219608 | 0.026269 | H | -0.549544 | 1.893254 | -1.725449 |
| C | 1.124582 | -0.005684 | 1.327335 | H | -2.894586 | 2.031481 | -1.740309 |
| C | -0.596642 | 0.101818 | -0.475209 | H | -3.073982 | 0.382005 | -2.311087 |
| C | -1.572921 | 0.654939 | 0.62704 | H | -3.183504 | -0.521043 | -0.139376 |
| C | -1.372519 | -0.161866 | 1.914092 | H | -0.260255 | 2.286415 | 1.305477 |
| C | 0.101333 | -0.278745 | 2.337009 | H | -1.431554 | 2.790969 | 0.084368 |
| C | -1.014466 | 0.903143 | -1.719614 | H | -1.955947 | 2.485149 | 1.738368 |
| C | -2.559618 | 1.001834 | -1.574575 | H | -0.785573 | -2.031133 | 0.09727 |
| C | -2.929411 | 0.543835 | -0.124816 | H | -1.75713 | -1.578605 | -1.317405 |
| C | -1.282621 | 2.139077 | 0.948554 | H | -0.000112 | -1.748233 | -1.450244 |
| C | -0.809911 | -1.406451 | -0.799109 | H | -3.935299 | 2.365038 | 0.403628 |
| O | 0.382492 | -0.60771 | 3.491964 | H | -4.505179 | -0.164055 | 2.038412 |
| C | 2.013701 | 0.530053 | -0.771187 | H | -5.45518 | 1.323647 | 2.185781 |
| C | -4.156318 | 1.289563 | 0.435815 | H | -3.73716 | 1.328079 | 2.584778 |
| C | -4.478244 | 0.921686 | 1.892078 | H | -6.16959 | 1.809769 | -0.119322 |
| C | -5.405761 | 1.100053 | -0.457809 | H | -5.174905 | 1.361496 | -1.49529 |
| C | -6.054249 | -0.298982 | -0.437171 | H | -6.25544 | -0.632422 | 0.58258 |
| C | -5.258063 | -1.352915 | -1.169386 | H | -7.018797 | -0.236619 | -0.955518 |
| O | -4.99811 | -2.435169 | -0.400371 | H | -4.495177 | -3.064962 | -0.953017 |
| O | -4.88875 | -1.276093 | -2.326124 | H | 4.144859 | 0.962676 | 2.812782 |
| C | 3.053854 | 0.956603 | 2.70902 | H | 2.613184 | 0.627326 | 3.653049 |
| C | 3.166174 | -1.407249 | 1.880625 | H | 2.724968 | 1.982328 | 2.515982 |
| C | 3.158131 | 0.494013 | 0.202944 | H | 4.26072 | -1.367633 | 1.915157 |
| C | 3.583628 | 2.46848 | -1.349988 | H | 2.818941 | -1.675514 | 2.882828 |
| C | 3.643132 | 3.97784 | -1.34129 | H | 3.974697 | 4.334405 | -0.358455 |
| O | 4.224699 | 1.794395 | -2.14951 | H | 4.324074 | 4.336772 | -2.115132 |
| C | 2.72485 | -2.534305 | 0.928602 | H | 2.640815 | 4.391348 | -1.507168 |
| C | 3.197285 | -2.415902 | -0.503909 | H | 3.120035 | -3.480435 | 1.319551 |
| O | 4.529886 | -2.223934 | -0.575717 | H | 1.637299 | -2.620913 | 0.901155 |
| O | 2.486571 | -2.504569 | -1.488215 | H | 4.722579 | -1.017811 | -2.261367 |
| C | 5.063223 | -1.990228 | -1.894334 | H | 6.146142 | -1.991055 | -1.774798 |
| H | 4.1451 | 0.184396 | -0.116784 | H | 4.74969 | -2.778242 | -2.581287 |
| H | 2.105256 | 0.248467 | -1.813297 |  |  |  |  |
| H | 2.137062 | 2.502906 | 0.311882 |  |  |  |  |
| H | -1.902682 | 0.267764 | 2.767534 |  |  |  |  |

## Uncropped images of western blot

Figure S82. Uncropped images of western blot in figure 5 of paper.


Figure S83. Uncropped images of western blot in figure S6 of Supplementary information.


Figure S84. Uncropped images of western blot in figure 6 of paper.

| A |  | C |  | $E$ |  | G |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| pAMPK ${ }^{\text {a }}$ | ---------------------- | pAMPK ${ }^{\text {a }}$ | [-------------------- | pAMPKa | [-------------------- | pAMPK ${ }^{\text {a }}$ | 「------------------------- |
| AMPKa | $\mid$ | AMPKa | -------------------- | AMPKa | ---------------------- | AMPKa | ------------------------- |
| AMPK $\beta 1$ AMPKß2 |  | AMPK $\beta 1$ AMPK $\beta 2$ |  | AMPK $\beta 1$ AMPKß2 |  | AMPKß1 AMPKß2 |  |
| pAMPKß1 | ----------------------- | pAMPKß1 | ----------------------- | pAMPK 31 | ---------------------- | pAMPKß1 | ------------------ |
| pULK1ser777 |  | pULK1ser777 | --------------------- | pULK1ser777 |  | pULK1ser777 | -1 |
| pULKser757 |  | pULKser757 | --\% | pULK1ser555 | ------------------ | pULK1ser555 |  |
| pULK1ser317 |  | pULKser317 | ------------------ | pULK1ser317 |  | pULKser317 |  |
| pULK1ser555 |  | pULK1ser555 |  | pULKser757 |  | pULKser757 |  |
| ULK1 |  | ULK1 |  | ULK1 | [----------------------- | ULK1 | ------------------------- |
| АСТВ | [---------------------- | ACTB | -------------------- | АСтв | ---------------------- | АСТВ | --------------------- |

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[^2]:    -- End Of Report --

[^3]:    .-- End Of Report -

